Total Time: 3:00 Hours (Selected Course Note: There are three sections in this paper he A B & VERSION : Prost an employed which A on the MCOs Answer Sheet only. Carrollack ball point or marker may be used for shading the relevant circle. · Please make sure that there is no cutting, over writing, erating, or multiple circles shading. Marks: 18 "Section-A" Time Allowed: 20 Minutes Q.1. Choose the correct option i.e. A,B,C, or D. When a dielectric medium is inserted between the two-point charges then the magnitude of electrostatic force Not affected Becomes zero (Increases 2. Electric field intensity due to an infinite sheet of charge is given by In Ohm's Law R = +, so by doubling the voltage, the resistance will (a) Remains the same (Half @ Four-times (A) Double 4. Which of the following is an example of wire-wound variable resistors? (i) Wheat stone bridge (Thermistors Potentio meter (c) Rheostats 5. Ampere's circuital law is given by the relation Β = μ₀ Β = μοΙ 6. One weber is equal to . N ⋅ A²/m 7. Which of the following quantities constant in step-up transformer? (Current (C) Voltage Heat What's the coefficient of mutual inductance, when the magnetic flux changes by 2×10^{-2} Wb, and change in current is 0.01A? (Zero @ 1/2 H 9. In pure capacitive circuit the power loss is (c) Infinite Minimum The peak value of alternating current is 5√2 A. The mean square value of current will be @ 2.5A 11. Which of the following is paramagnetic material? (Bismuth Copper Antimony (c) Zinc 12. In transistor, there are PN junctions. (Five (A) Four ① Three 13. The laws of physics are the same in all inertial reference frames. It is stated by (c) Galileo (Thomson (A) Bohr 14. The maximum K.E of emitted photoelectron depends upon (Temperature of the Partitional light Frequency of the incident light metal surface 15. According to Bohr's theory of hydrogets storm the hadit of the first orbit is @13(4)x(10940) ⑤ 5.3 × 10⁻⁹ m 13.6 × 10⁻⁹ m ⊚ −13.6 eV ⊕ −1.51 eV 17. In terms of energy 1 a.m. u = @ 938 Me @ 937 MeV. @ 931 MeV 18. A radioactive substance has a half-life of four months. 3-fourth of the substance will decay in © 8 months (©) 6 months

12 months

16 months

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