

Revision Notes Class 12

Physics

Chapter 14 - Semiconductor Electronics

1. ENERGY BANDS IN SOLIDS

In case of solids, there are single energy levels where the atoms are arranged in a systematic space lattice and therefore the atom is highly influenced by the neighboring atoms for a single isolated atom. The closeness of atoms will be resulting in the intermixing of electrons of the neighboring atoms of course, for the valence electrons, not strongly bounded by the nucleus, in the outermost shells. Because of the intermixing the number of permissible energy levels will be increased or there will be significant variations in the energy levels. Therefore, instead of single energy levels related with the single atom in the case of a solid, there will be bands of energy levels.

1.1 Conduction Band, Valence Band & Forbidden Energy Gap

The band created by a series of energy levels including the valence electrons can be defined as valence band. The valence band can be explained as a band which is inhabited by the valence electrons or a band which is having highest inhabited band energy. The conduction band can be explained as the least unfilled energy band. The forbidden energy gap can be defined as the separation between conduction band and valence band. There will be no permitted energy state in this gap and therefore electron can't occupy in the forbidden energy gap.

1.2 Conductors, Semiconductors and Insulators

Conductors are the materials which can conduct the charge carriers easily. Insulators are the materials in which the free flow of charge carriers is not possible. And semiconductors are the materials which is having conductivity in between insulators and conductors. According to the forbidden band, the insulators, semiconductors and conductors can be explained as mentioned below:

1.2.1 Insulators

The forbidden energy band will be very wide in case of insulators. Because of this information electrons will not be jumped from valence band to conduction

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