Prentice Hall Physical Science Chapter 4 Atomic Structure

4.1 Studying Atoms

- A. Ancient Greek Models of Atoms
 - Democritus believed all matter was made of extremely small particles that could not be divided
 - he called the particles atoms (from the Greek atomos meaning indivisible)
 - he thought there were different types of atoms with different properties (liquids had round, smooth atoms and solids were rough and prickly)
- B. Dulton's Atomic Theory
 - be proved atoms exist by discovering that compounds have a fixed composition
 - Dalton's Theory:
 - all elements are composed of atoms
 - all atoms of the same element have the same mass, and atoms of different elements have different masses.
 - compounds contain atoms of more than one element
 - in a particular compound, atoms of different element always combine in the same way
 - he pictured atoms as solid spheres
 - in time it was found that not all of this was completely correct
- C. Thomson's Model of the Atom
 - some objects have a positive or negative charge and objects with the same charge repel and objects with opposite charges attract
 - Thomson used an electric current to learn more about atoms
 - he passed an electric current through a gas and a glowing beam appeared
 - the beam was repelled by a negatively charged metal plate (like charges repel and opposite charges attract)
 - Thomson concluded that the beam most be made of negatively charged particles, smaller than atoms.
 - since the atom was known to be electrically neutral, he developed a model in which negative charges were
 - evenly scattered throughout an atom filled with a positively charged pudding like material called the "Plum Pudding" model (or chocolate chip cookie dough)
 - called the Plum Pudding in
- D. Rutherford's Atomic Theory
 - Rutherford discovered positively charged, fast moving particles called alpha particles
 - be shot some of these particles a piece of gold foil, thinking they would go straight through
 - when he actually performed the experiment, he found that the positively charged alpha particles were greatly deflected by the gold (figure 7, p. 104)
 - he determined that the alpha particles must have come very close to another positively charged particle causing them to be repelled and put off course
 - this led him to believe that the positive charge in an atom is not evenly distributed, but concentrated in a very small central area he called the nucleus
 - his model of the atom is a small, positively charged nucleus surrounded by negatively charged particle

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assessments in the wide range of multilingual contexts in Africa **An Introduction to Physical Science** James T. Shipman, Jerry D. Wilson, Aaron W. Todd, 2002-07-08 This text presents a survey of the physical sciences physics chemistry astronomy meteorology and geology for non science majors Topics are treated both descriptively and quantitatively providing flexibility for instructors who wish to emphasize a highly descriptive approach a highly quantitative approach or any Handbook of Research on Science Learning Progressions Hui Jin, Duanli Yan, Joseph Krajcik, 2024-07-30 Gathering contributions from leading scholars around the world this handbook offers a comprehensive resource on the most recent advances in research surrounding the theories methodologies and applications of science learning progressions Researchers and educators have used learning progressions to guide the design and alignment of curriculum instruction and assessment and to help students learn scientific knowledge and practices in a coherent and connected way across multiple years This handbook lays out the development and current state of research in this field across four sections learning progression theories and methodologies learning progressions to promote student learning teachers learning and use of learning progressions and new technology in learning progression research Featuring internationally recognized experts in learning progression research as well as up and coming voices the Handbook of Research on Science Learning Progressions offers a defining new resource for researchers teachers and teacher educators The GED For Dummies® Murray Shukyn, Dale E. and curriculum and assessment developers in science education Shuttleworth, 2010-04-20 Get the skills and know how you need to pass the GED test Earning a GED can provide you with an advantage over other job and education candidates and the confidence to take the next step The GED For Dummies 2nd Edition gives you fresh and relevant example questions from the GED and even more hands on training in each of the 5 subject areas to help you maximize your success and earn a passing score Features 2 full practice tests in each of the 5 subject areas with detailed walk throughs and explanations for every solution Offers advice on test preparation from registering and studying effectively to managing your time during the exam Improve your job and education prospects now by studying for the GED with this easy to follow proven guide Arun Deep's Self-Help to ICSE Geography Class 10: 2024-25 Edition (Based on Latest ICSE Syllabus) Jine Laxmi, 2024-03-01 Self Help to I C S E Geography Class 10 has been written keeping in mind the needs of students studying in 10th I C S E This book has been made in such a way that students will be fully guided to prepare for the exam in the most effective manner securing higher grades. The purpose of this book is to aid any I C S E student to achieve the best possible grade in the exam This book will give you support during the course as well as advice you on revision and preparation for the exam itself. The material is presented in a clear concise form and there are ample questions for practice KEY FEATURES Chapter At a glance It contains the necessary study material well supported by Definitions Facts Figures Flow chart etc Solved Questions The condensed version is followed by Solved Questions and Map based Picture based questions along with their Answers This book also includes the Answers to the

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