



basic education

Department:  
Basic Education  
REPUBLIC OF SOUTH AFRICA

# PHYSICAL SCIENCES

## GRADE 12

## JUNE EXAMS- MEMO

# 2023

QUESTION 7 (start on a new page.)

- 7.1 Define an acid according to the Brønsted-Lowry theory.
- 7.2 The table below shows the ionisation constants ( $K_a$ ) of two acids of equal concentrations.

NAME	FORMULA	$K_a$
Sulphuric acid	$\text{H}_2\text{SO}_4$	$1,2 \times 10^{-2}$
Sulphurous acid	$\text{H}_2\text{SO}_3$	$1,54 \times 10^{-2}$

- 7.2.1 Which ONE of these acids will have a higher molar conductivity? Give a reason for this answer.
- 7.2.2 Write down a chemical reaction to show the ionisation of sulphuric acid in water.

A	B
C $\text{CH}_3\text{Cl}$	D Propanoic acid
E Propanol	F $\text{C}_2\text{H}_5\text{Cl}$

2011

- 3.3.4 Explain the shape of the graph and what happened after  $t_0$ . (2)

- 3.4 The following Maxwell-Boltzmann distribution graph was obtained at a low temperature.



- 3.4.1 Copy the graph given above in your ANSWER BOOK. Use a dotted line and indicate on the graph how the distribution would change if a HIGHER TEMPERATURE.

20

# Physical Sciences Paper 2 June Examination Of Eastern Cape

**Sir Norman Lockyer**



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### **Table of Contents Physical Sciences Paper 2 June Examination Of Eastern Cape**

1. Understanding the eBook Physical Sciences Paper 2 June Examination Of Eastern Cape
  - The Rise of Digital Reading Physical Sciences Paper 2 June Examination Of Eastern Cape
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Sciences Paper 2 June Examination Of Eastern Cape
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Sciences Paper 2 June Examination Of Eastern Cape
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Sciences Paper 2 June Examination Of Eastern Cape
  - Personalized Recommendations
  - Physical Sciences Paper 2 June Examination Of Eastern Cape User Reviews and Ratings
  - Physical Sciences Paper 2 June Examination Of Eastern Cape and Bestseller Lists

5. Accessing Physical Sciences Paper 2 June Examination Of Eastern Cape Free and Paid eBooks
  - Physical Sciences Paper 2 June Examination Of Eastern Cape Public Domain eBooks
  - Physical Sciences Paper 2 June Examination Of Eastern Cape eBook Subscription Services
  - Physical Sciences Paper 2 June Examination Of Eastern Cape Budget-Friendly Options
6. Navigating Physical Sciences Paper 2 June Examination Of Eastern Cape eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Sciences Paper 2 June Examination Of Eastern Cape Compatibility with Devices
  - Physical Sciences Paper 2 June Examination Of Eastern Cape Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Sciences Paper 2 June Examination Of Eastern Cape
  - Highlighting and Note-Taking Physical Sciences Paper 2 June Examination Of Eastern Cape
  - Interactive Elements Physical Sciences Paper 2 June Examination Of Eastern Cape
8. Staying Engaged with Physical Sciences Paper 2 June Examination Of Eastern Cape
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Sciences Paper 2 June Examination Of Eastern Cape
9. Balancing eBooks and Physical Books Physical Sciences Paper 2 June Examination Of Eastern Cape
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Sciences Paper 2 June Examination Of Eastern Cape
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Sciences Paper 2 June Examination Of Eastern Cape
  - Setting Reading Goals Physical Sciences Paper 2 June Examination Of Eastern Cape
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Sciences Paper 2 June Examination Of Eastern Cape
  - Fact-Checking eBook Content of Physical Sciences Paper 2 June Examination Of Eastern Cape
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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