NAME	Period	Date	
	Predicting React	ion types	DOSD, DD, CORNE
equation for each of the	ction (or NR for no reaction.) complete to following. Make sure to include phase etal and hydrochloric acid →	symbols K VX UK	he state of master
Para Laterum nydro	(5) + \\\(\lambda\)\(\alpha\)\(\alpha\)\(\alpha\)		
NR 3. Aluminum and	(0H) ≥ (aq) + 2HNO 5 (aq)	> Ca(NO)2 +2420
4. Mercury(III) an	(5) + Mg(5) X>		moveus solids of both fund to
DO 5. Zinc chloride a	(L) + O ₂ (g) → RI nd hydrogen sulfide →	1g O (s)	
	Cl2 (aq) + H2S (aq) -	\rightarrow ZnS(s)	+2HCI (of)
D 7. Sodium chlora 2 N	$CI(S) + Nb_1Nb_3(aq)$ the is heated \Rightarrow $a_1CIO_3(s) \xrightarrow{\Delta} > 2Na_2CV_{aq}$		be a prompitale, 40,
DD 9. Sodium bromi	and sodium chromate \Rightarrow $O(NO_3)_2 (O_2)_1 + NO_2O_r$ de and silver nitrate \Rightarrow	O4 (ag)	-> Ba Cr 04(5) +2NaNO3
	Br (q) + Ag NO ₃ (aq) — sphate and aluminum sulfate →		s) + Na NO3 (aq)
D 11. Zinc carbonal	$3 (PD4)_2 (S) + Al_2 (S)$ te is heated \Rightarrow	(PA)E(HO	> 2 M PO4(9+3CaSO4(5)
	$CO_3(s) \xrightarrow{\triangle} ZnO(s)$		
S,C 13. Hydrogen and) + F2 (9) -> 2KI	= (s) mag ? ,	
NR 14. Lead(II) and p	(g) + Oz(g) -> 21/2(0 (2)	pagastan + mulate at the
	s) + KC10, (aa) -	*>	
6 Na ((s) +2H2 PO4 (a) ->	2 No 3 PO4 (09)	+ 3Ha(g)

Writing Equations Predicting Reactions Answer Key

Morris Hein, Susan Arena

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