

Periodic Law

OBJECTIVES

- To observe and investigate the periodic trends for reactivity, density, pH and solubility within groups 1, 2 and 14 on the periodic table of the elements.

INTRODUCTION

When the elements are arranged in order of increasing atomic number, there is a periodic occurrence of properties that leads to the grouping of elements on the periodic table. Elements in vertical columns form groups or families that are characterized by marked similarities in physical and chemical properties. These similarities are due to the fact that all of the elements within a group have the same number of valence electrons. Within a group, periodic trends in certain properties can be observed. The properties of density and solubility, for example, are known to vary regularly within groups of elements.

PROCEDURE

PART A

The teacher will demonstrate the reactivity of three elements (Lithium, Sodium and Potassium) in group 1, by adding a small piece of the metal to water. Observe data sheet and record observations and rank the three elements in order of reactivity.

Element	Reactivity	Observations
Li		
Na		
K		

PART B

- Pour about 5 mL of distilled water into a clean dry test tube.
- FRONT LAB TABLE:** Remove the pH probe with caution from the water bottle and insert the pH probe into the distilled water. Note the pH on the "Lab Quest". Remove the pH probe and store with the water bottle.
- FRONT LAB TABLE:** Add a few drops of phenolphthalein solution to the reaction tube. Phenolphthalein solution turns pink in the presence of a base.

The Periodic Law Lab Answers

Harold Clark Metcalfe



The Periodic Law Lab Answers:

Laboratory Manual for Principles of General Chemistry J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don't write experiments A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition's experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment's Report Sheet to provide a space for recording observations and data during the experiment Continued

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