

# Question 5

1. A chemical reaction involving protons and hydroxide ions is shown below. Calculate the equilibrium constant  $K_a$  for the reaction at 25°C. The concentration of the products is 0.1 M and the concentration of the reactants is 0.1 M.

$$K_a = \frac{[H^+][A^-]}{[HA]}$$

2. Calculate the pH of a 0.1 M solution of  $H_2PO_4^-$ . Give your answer to two decimal places.

$$pH = -\log \sqrt{(4.0 \times 10^{-8})(0.1)} = 0.075$$

$$-\log(0.0075) = 2.12$$

$$pH = 2.12$$

3. The chemical reaction below is a buffer reaction. Calculate the pH of a 0.1 M solution of  $H_2PO_4^-$  with 0.1 M of  $HPO_4^{2-}$  added. (Ka =  $4.0 \times 10^{-8}$ ).

4. Calculate the pH of the buffer solution.

Give your answer to two decimal places.

$$0.1 \times \frac{4.0 \times 10^{-8}}{0.1} = 0.04$$

$$0.045 \times \frac{4.0 \times 10^{-8}}{0.1} = 0.05$$

$$(4.6 \times 10^{-8}) \times \frac{0.04}{0.05} = 0.003752$$

$$-\log(0.003752) = 2.43$$

$$2.43$$

5. A chemical reaction involving protons and hydroxide ions is shown below. Calculate the equilibrium constant  $K_a$  for the reaction at 25°C. The concentration of the products is 0.1 M and the concentration of the reactants is 0.1 M.

6. A chemical reaction involving protons and hydroxide ions is shown below. Calculate the equilibrium constant  $K_a$  for the reaction at 25°C. The concentration of the products is 0.1 M and the concentration of the reactants is 0.1 M.

$$pH = -\log \sqrt{(4.0 \times 10^{-8})(0.1)} = 0.075$$

$$-\log(0.0075) = 2.12$$

7. A chemical reaction involving protons and hydroxide ions is shown below. Calculate the equilibrium constant  $K_a$  for the reaction at 25°C. The concentration of the products is 0.1 M and the concentration of the reactants is 0.1 M.

$$pH = -\log \sqrt{(4.0 \times 10^{-8})(0.1)} = 0.075$$

8. A chemical reaction involving protons and hydroxide ions is shown below. Calculate the equilibrium constant  $K_a$  for the reaction at 25°C. The concentration of the products is 0.1 M and the concentration of the reactants is 0.1 M.

# Ocr Chemistry 2014 Unofficial Mark Scheme

**Arturo Cuomo**



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