

সিলেট বোর্ড

বহুনির্বাচনি অধীক্ষা

তদারকন

প্রথম পত্র

[২০২৩ সালের সিলেটবাস অধ্যায়ী]

সময়—২৫ মিনিট

পূর্ণমান—২৫

বিষয় কোড :

1 7 6

সেট :

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[বিশেষ দ্রষ্টব্য : সরবরাহকৃত বহুনির্বাচনি অধীক্ষার উত্তরপত্রে প্রশ্নের ত্রিমিক নম্বরের বিপরীতে প্রদত্ত বর্ণসম্বলিত বৃত্তসমূহ হতে সঠিক/সর্বোৎকৃষ্ট উত্তরের বৃত্তটি বলা পরেই কলাম দ্বারা সম্পূর্ণ করতে কর। প্রতিটি প্রশ্নের মান ১।]

প্রশ্নপত্রে কোনো প্রকার দাণ্ড/চিহ্ন দেয়া যাবে না।

- ১। চিকিৎসা বিজ্ঞানে ডিজিগবেষণিতে কোনটি ব্যবহার করা হয়?

(ক) X-ray
(খ) IR
(গ) MRI
(ঘ) UV

উল্লিখিত থেকে ২ এবং ৩ নং প্রশ্নের উত্তর দাও :

মৌল	বহির্গতরের ইলেকট্রন বিন্যাস	সকল কেসে n এর মান 2
A	$(n-1)s^1$	
B	ns^2np^4	
C	$(n+1)s^2(n+1)p^4$	

A, B এবং C প্রদত্ত মৌলের প্রতীক নয়।

- ২। A_2B ও A_2C যৌত অবস্থার ত্রিমাত্রিক ক্রম—

(ক) আয়নিক বন্ধন
(খ) সমতাপীয় বন্ধন
(গ) হাইড্রোজেন বন্ধন
(ঘ) সলিড বন্ধন

- ৩। A_2B যৌলের ক্ষেত্রে—

I. sp^3 সংকরণ ঘটে
II. বন্ধন কোণ 104.5°
III. $(L_p - L_p)$ বিকর্ষণ বিদ্যমান

নিচের কোনটি সঠিক?

(ক) I ও II
(খ) II ও III
(গ) I ও III
(ঘ) I, II ও III

- ৪। নিচের মৌলগুলোর ১ম আয়নীকরণ বিভবের সঠিক ক্রম কোনটি?

(ক) $Be > B > N > O$
(খ) $N > O > Be > B$
(গ) $O > N > B > Be$
(ঘ) $B > Be > N > O$

- ৫। হেলিনের প্রকৃত ভর কত?

(ক) $1.60 \times 10^{-24} g$
(খ) $1.66 \times 10^{-24} g$
(গ) $1.673 \times 10^{-24} g$
(ঘ) $1.675 \times 10^{-24} g$

- ৬। কোনটি প্রথম অক্সাইড?

(ক) NO
(খ) N_2O_5
(গ) N_2O_3
(ঘ) NO_2

- ৭। S-গ্রক মৌলের সংখ্যা কতটি?

(ক) 7
(খ) 12
(গ) 14
(ঘ) 16

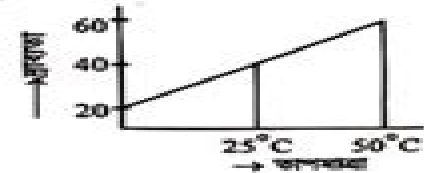
- ৮। পরমাণুর ৩য় শক্তিস্তরে ঘূর্ণিত ইলেকট্রনের কৌণিক ভরবেগ এর মান কোনটি?

(ক) $\frac{nh}{2\pi}$
(খ) $\frac{3h}{2\pi}$
(গ) $\frac{2h}{4\pi}$
(ঘ) $\frac{9h}{2\pi}$

- ৯। কোনটি উভয়দিকী অক্সাইড?

(ক) Al_2O_3
(খ) Na_2O_2
(গ) KO_2
(ঘ) Fe_2O_3

উল্লিখিত হতে ১০ এবং ১১ নং প্রশ্নের উত্তর দাও :



- ১০। 200g NH_4Cl এর সম্পূর্ণ দ্রবককে $50^\circ C$ থেকে $25^\circ C$ তাপমাত্রায় ঠাণ্ডা করলে কত গ্রাম NH_4Cl ক্রিস্টালিত হবে?

(ক) 25
(খ) 50
(গ) 75
(ঘ) 90

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C. M. Roebuck



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Excel HSC & Preliminary Senior Science Jennifer Hill, 2011 This comprehensive study guide covers the complete HSC Preliminary Senior Science course and has been specifically created to maximise exam success This guide has been designed to meet all study needs providing up to date information in an easy to use format The sample HSC Exam has been updated for the new format Excel HSC Preliminary Senior Science contains an introductory section including how to use the book and an explanation of the new course helpful study and exam techniques comprehensive coverage of the entire Preliminary and HSC courses hundreds of diagrams to aid understanding icons and boxes to highlight key concepts and assessment skills including laboratory and field work checklists of key terms end of chapter revision questions with fully explained answers a trial HSC style exam with answers and explanations a glossary of key terms useful websites highlighted throughout **Excel Chemistry** C. M. Roebuck, 1995 **HSC Chemistry Core Study Material**, 198? *HSC Chemistry*

Marilyn Schell, Margaret Hogan, 2007-09-01 In this book you will find nearly 600 typical multiple choice examination questions nearly 600 extension questions plus summaries for the Year 12 Chemistry course Answers to all questions are provided Questions follow the dot points in the Board of Studies syllabus and focus on the three core topics plus the option topic Shipwrecks Corrosion and Conservation **HSC questions and answers 2 unit chemistry revision guide**, 1992

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