

## Chapter – 8

### B. Very Short Answer Type Questions:

1. Write the molecular formula of aluminium oxide.

Ans:  $\text{Al}_2\text{O}_3$

2. The elements present in sodium bicarbonate are sodium, carbon, \_\_\_\_\_ and oxygen.

Ans: Hydrogen

3. Which one has more mass 1 mole of hydrogen gas or 0.20 mole of nitrogen gas?

Ans: 0.20 mole of nitrogen gas

4. Which one has more number of molecules 1 mole of  $\text{N}_2$  or 1 mole of  $\text{O}_2$ ?

Ans: Same. Because 1 mole of any substance always contain  $6.022 \times 10^{23}$  of that substance.

5. What is the value of Avogadro's constant?

Ans:  $6.022 \times 10^{23}$

6. What is the valency of carbon in (a)  $\text{CCl}_4$  and (b)  $\text{CO}_2$ ?

Ans: (a)  $\text{CCl}_4$  – 4, (b)  $\text{CO}_2$  – 4

7. If the atomic symbol of Cobalt is Co, then what is the atomic symbol of Copper?

Ans: Cu

8. What is the number of moles in 18g of water?

Ans: 1

9. What is the mass of 1 mole of water?

Ans: 18g

### C. Short Answer Questions Type - I:

1. State the Law of constant proportion and name the scientist who proposed this law?

Ans: Law of constant proportion states that,

A chemical compound always contains the same element combined together in the same definite ratio by mass.

This law was proposed by Joseph L. Proust.

# Sciencechapter 8 Review And Assessment Answer

**Kaplan Test Prep**



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