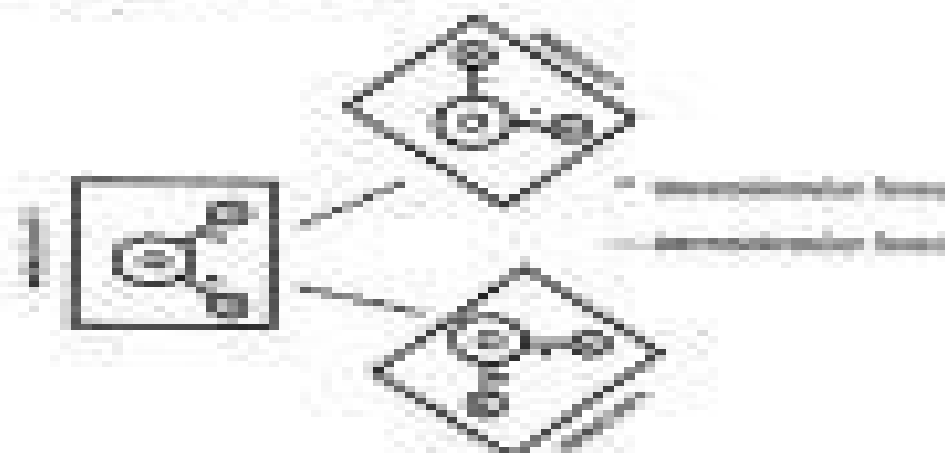


# POGIL: Intermolecular Forces

## Model 1: What is an Intermolecular Force?

As you have learned, matter is made up of discrete particles called atoms which chemically combine to form molecules. Molecules do not exist as independent units; in fact, groups of molecules "stick together" in order to form liquids and solids. The forces that hold groups of molecules together are Intermolecular Forces. Without intermolecular forces, the world as we know it would not be the same.

Figure 1: Intermolecular and Intramolecular Forces



## Critical Thinking Questions

1. What specific molecule is represented inside each box in Figure 1?  
 $\text{H}_2\text{O}$
2. In relation to the box for molecule 1, where do the intermolecular forces exist in Figure 1—inside the boxes or outside the boxes?  
 $\text{outside the boxes}$
3. Based on the intermolecular forces for molecule 1, draw similar patterns (?) for the intermolecular forces of the (diagram the molecules 2 and 3).
4. In relation to the molecule, where do intramolecular forces tend to occur—within the molecule or outside of the molecule?  
 $\text{within the molecule}$
5. Draw intermolecular forces under in Figure 1. Where are they positioned relative to the molecules—within the molecules or between the molecules?  
 $\text{between the molecules}$

6. State the difference between intermolecular and intramolecular forces in terms of where they exist on the molecular level.

intermolecular = happens  $\text{between}$  separate molecules, like boiling water, or melting something  
 intramolecular = happens  $\text{within}$  each molecule, like taking apart a molecule  
 (happens on the inside-out)

# Pogil Intermolecular Forces Key

**Richard Samuel Moog**



## **Pogil Intermolecular Forces Key:**

**Process Oriented Guided Inquiry Learning (POGIL)** Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes

**Intermolecular Forces** , **Intermolecular Forces** Pierre L. Huyskens, Werner A.P. Luck, Therese Zeegers-Huyskens, 2012-12-06 The study of intermolecular forces began over one hundred years ago in 1873 with the famous thesis of van der Waals In recent decades knowledge of this field has expanded due to intensive research into both its theoretical and the experimental aspects This is particularly true for the type of very strong cohesive force stressed in 1920 by Latimer and Rodebush the hydrogen bond a phenomenon already outlined in 1912 by Moore and Winemill Hydrogen bonds exert a profound influence on most of the physical and chemical properties of the materials in which they are formed Not only do they govern viscosity and electrical conductivity they also intervene in the chemical reaction path which determines the kinetics of chemical processes The properties of chemical substances depend to a large extent on intermolecular forces In spite of this fundamental fact too little attention is given to these properties both in research and in university teaching For instance in the field of pharmaceutical research about 13000 compounds need to be studied in order to find a single new product that can be successfully marketed The recognition of the need to optimize industrial research efficiency has led to a growing interest in promoting the study of inter molecular forces Rising salary costs in industry have encouraged an interest in theoretical ideas which will lead to tailor made materials

Intermolecular Forces Geoffrey C. Maitland, 1981

**Intermolecular Forces** A. Pullman, 2014-01-15

**Intermolecular Forces** A. Pullman, 2013-06-29 Proceedings of the 14th Jerusalem Symposium on Quantum Chemistry and Biochemistry Jerusalem Israel April 13 16 1981

**Theory of Intermolecular Forces** Henry Margenau, N. R. Kestner, 1969

**Intermolecular Forces** Taro Kihara, 1976

**Intermolecular and Surface Forces** Jacob N. Israelachvili, 2011-07-22 Intermolecular and Surface Forces describes the role of various intermolecular and interparticle forces in determining the properties of simple systems such as gases liquids and solids with a special focus on more complex colloidal polymeric and biological systems The book provides a thorough foundation in theories and concepts of intermolecular forces allowing researchers and students to recognize which forces are important in any particular system as well as how to control these forces This third edition is expanded into three sections and contains five new chapters over the previous edition Starts from the basics and builds up to more complex systems Covers all aspects of intermolecular and interparticle forces both at the fundamental and applied levels Multidisciplinary approach bringing together and unifying phenomena from different fields This new edition has an expanded Part III and new chapters on non equilibrium dynamic interactions and tribology friction forces

Intermolecular Forces and Clusters , 2005

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