

4. Match each description below to one of the reactions sets (A, B, C or D) from Model 1.

<u>C</u>	Ionic compounds dissolved in water switch partners.
<u>D</u>	One compound breaks into elements or smaller compounds.
<u>A</u>	Two or more elements or compounds combine to form one product.
<u>B</u>	Part of an ionic compound is removed and replaced by a new element.

5. Define the following terms as they are commonly used in the English language.

Synthesis—Combination or Composition

Decomposition—process of decay

Replacement—replacing someone or something

6. The four sets of chemical reactions shown in Model 1 have the following general names. Discuss within your group which name belongs to which set of chemical reactions. Write the name in the appropriate place in Model 1.

Single Replacement Reaction C

Synthesis Reaction A

Double Replacement Reaction D

Decomposition Reaction B

7. Can two elements be used as reactants for a synthesis reactions? If yes, give at least one example from Model 1 to support your answer.



8. Can two compounds be used as reactants for a synthesis reactions? If yes, give at least one example from Model 1 to support your answer.



9. What types of substances (elements or compounds) are seen in the products of decomposition reactions? Use examples from Model 1 to support your answer.

Both elements (O_2) and compounds (CO_2)

10. In single replacement reactions, do any of the atoms change their charge? If yes, use an example from Model 1 to describe the changes that take place.

Yes, FeCl_3 turns into just Fe.

11. In double replacement reactions, do any of the atoms change their charge? If yes, use an example from Model 1 to describe the changes that take place.

Yes, NaCl to AgCl

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SB Merriam

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