#### Chapter 6 Rates of Chemical Reactions

#### Practice Problems

#### Problem 1

The formation reaction of ammonia is represented by the following equation:

N<sub>high</sub> + 3H<sub>high</sub> → 2NH<sub>high</sub>

Ammonia gas is produced at a rate of 4.0 × 10 3 mol/(L·s).

(a) What will be the rates at which the reactants are consumed?

(b) Assume that the reaction can proceed to completion. If the reaction begins with  $1.0 \times 10^2$  mol of  $N_{2ij}$  and  $2.0 \times 10^2$  mol of  $H_{2ij0}$  how much  $N_{2ij0}$  will remain when all  $H_{2ij0}$  is used up?

(c) If the reactants in (b) are placed in a 100 L reaction vessel and the reaction is allowed to proceed to completion, how long will it take?

#### Problem 2

For the reaction  $5O_{2igh} + 4NH_{3igh} \rightarrow 6H_2O_{igh} + 4NO_{igh}$ , the initial rate of production of  $H_2O$  is  $2.64 \times 10^{-6} \text{ mol/}(L\text{-s})$ . What is the corresponding rate of disappearance of  $O_2$ ?

#### Problem 3

The rate law equation for the reaction  $I_2 + (C_2H_3)_2(NH)_2 \rightarrow (C_2H_3)_2N_2 + 2HI$  is given below. Rate =  $kI(C_3H_3)_3(NH)_3I$ 

The initial rate of reaction is 0.250 mol/L when the concentration of (C<sub>2</sub>H<sub>2</sub>)<sub>2</sub>(NH)<sub>2</sub> is 30.0 mol/(L·s). What is the corresponding rate at which HI is formed if the initial concentration of (C<sub>2</sub>H<sub>2</sub>)<sub>2</sub>(NH)<sub>2</sub> is 0.150 mol/L?

#### Problem 4

An organic reaction between complex molecules is represented by the following reaction.

 $2A + 2B \rightarrow C + 2D$ 

The rate of reaction is found to obey a rate law equation. Use the rate data below to answer the questions.

Trial	Initial Concentration (mol/L)		Initial Rate mol/(L-s)
	IAI	[8]	
1	1.0	1.0	$2.1 \times 10^{-1}$
2	4.0	1.0	8.3 × 10 <sup>-1</sup>
3	1.0	2.0	8.4 × 10 <sup>-1</sup>

(a) What is the overall reaction order of this reaction?

(b) Calculate the value of the rate constant.

(c) What will be the effect upon the rate of reaction if the concentrations of both A and B are doubled?

# Rates Of Reaction 18 Practice Problems Worksheey

**Viktor Korobov, Valery Ochkov** 

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