7-1 Review and Reinforcement

Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative positive crisscross noble gas electrons anion

octet ions cation electron configuration polyatomic empirical formula

- An ionic bond is an attraction between oppositely charged —ions
- Anions have a -negative ----- charge.
- An atom becomes an ion by losing or gaining ____electrons
- The _octet __rule states that atoms tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
- When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) _noble gas
- 6. The atoms that make up **polyatomic** _ions are bonded together by covalent bonds.
- The _crisscross __method can be used to write the formula for an ionic compound.
- 8. The _empirical formula of a compound denotes the ratio of ions in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

- losing . 9. Calcium becomes a monatomic cation by gaining two electrons.
- _ true _____ 10. A cation has a positive charge.
- true . 11. A binary ionic compound contains only gag kind of cation and one kind of anion.
- _polyatomic __ 12. Monatomic ions consist of more than one atom.
- __seven _____13. The Lewis dot diagram for chlorine, a group 7A element, has aix electrons.
- __true (released) _ 14. A great deal of energy is produced when an ionic compound is formed.

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David Baud

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- 1. Understanding the eBook Study Guide And Reinforcement Answer Key Grade 8
 - The Rise of Digital Reading Study Guide And Reinforcement Answer Key Grade 8
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Study Guide And Reinforcement Answer Key Grade 8
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide And Reinforcement Answer Key Grade 8
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Study Guide And Reinforcement Answer Key Grade 8
 - Personalized Recommendations
 - Study Guide And Reinforcement Answer Key Grade 8 User Reviews and Ratings

- Study Guide And Reinforcement Answer Key Grade 8 and Bestseller Lists
- 5. Accessing Study Guide And Reinforcement Answer Key Grade 8 Free and Paid eBooks
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 - Study Guide And Reinforcement Answer Key Grade 8 Budget-Friendly Options
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 - Study Guide And Reinforcement Answer Key Grade 8 Enhanced eBook Features
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 - Highlighting and Note-Taking Study Guide And Reinforcement Answer Key Grade 8
 - o Interactive Elements Study Guide And Reinforcement Answer Key Grade 8
- 8. Staying Engaged with Study Guide And Reinforcement Answer Key Grade 8
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide And Reinforcement Answer Key Grade 8
- 9. Balancing eBooks and Physical Books Study Guide And Reinforcement Answer Key Grade 8
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Study Guide And Reinforcement Answer Key Grade 8
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Study Guide And Reinforcement Answer Key Grade 8
 - Setting Reading Goals Study Guide And Reinforcement Answer Key Grade 8
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Study Guide And Reinforcement Answer Key Grade 8
 - Fact-Checking eBook Content of Study Guide And Reinforcement Answer Key Grade 8
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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