

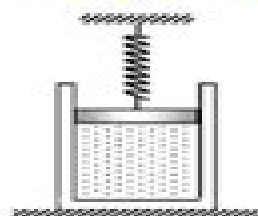
$$Q_C = Q_R + w$$

$$235,7 = 210 + w$$

$$w = 25,7 \text{ J}$$

RPTA.: E

14. En la figura se muestra un recipiente y un resorte de rigidez 50 N/m que está sin deformar, unido a un pistón de 1 kg. El recipiente tiene una capacidad calorífica 5 J/°C y contiene 3 kg de un gas combustible cuyo poder calorífico es 50 J/kg. Si el gas explota y los residuos de la combustión incrementan su energía interna en 30 J y la temperatura del sistema se eleva en 10°C, calcule la deformación del resorte. El pistón tiene una sección de 0,5 cm². Desprecie la fricción.



- A) 0,2 m
B) 0,5 m
C) 0,6 m
D) 0,8 m
E) 1,0 m

RESOLUCIÓN

Recipiente:

$$Q_R = 5 \frac{\text{J}}{^\circ\text{C}} \times 10^\circ\text{C} \Rightarrow Q_R = 50 \text{ J}$$

$$\text{Gases: } Q_T = Q_{\text{Gas}} + Q_R$$

$$(50 \text{ J/kg}) \times (3 \text{ kg}) = Q_{\text{Gas}} + 50$$

$$Q_{\text{Gas}} = 100 \text{ J}$$

$$\left. \begin{aligned} Q_{\text{Gas}} &= w + \Delta U \\ 100 &= w + 30 \end{aligned} \right\} W = 70 \text{ J}$$

$$w = (F_E + F_{\text{peso}}) \times$$

$$70 = (50x + 1 \times 10) \times$$

$$5x^2 + x - 7 = 0$$

$$x = 1,087 \text{ m}$$

RPTA.: E

15. La eficiencia teórica más alta de un motor de gasolina, basado en el ciclo

de Carnot, es de 25 %. Si este motor expulsa los gases a la atmósfera a una temperatura de 27 °C, ¿cuál es la temperatura en el cilindro inmediatamente después de la combustión de la gasolina?

- A) 127 °C B) 135 °C C) 140 °C
D) 180 °C E) 200 °C

RESOLUCIÓN

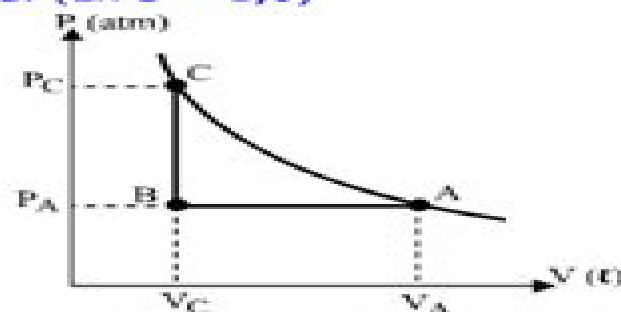
$$n = 1 - \frac{T_f}{T_c}$$

$$\frac{1}{4} = 1 - \frac{300}{T_c} \Rightarrow T_c = 400 \text{ K}$$

$$T_c = 127^\circ\text{C}$$

RPTA.: A

16. Un gas ideal se comprime lentamente a una presión constante de 2 atm, de 10 litros hasta 2 litros. En este proceso, algo de calor sale y la temperatura desciende. A continuación se agrega calor al gas, manteniendo constante el volumen, y se dejan aumentar la presión y la temperatura. Calcule el flujo de calor total hacia el gas. El proceso se muestra en la figura como el trayecto ABC. (Ln 5 = 1,6)



- A) - 1 000 J B) - 1 200 J
C) - 1 600 J D) + 1 200 J
E) + 1 600 J

RESOLUCIÓN

Proceso (A) → (B) ; Isobárico

Termodinamica Problemas Resueltos

Y Pai

A decorative red circular graphic with a gradient, appearing as a semi-circle or a stylized arrow pointing to the right, located to the right of the author's name.

Termodinamica Problemas Resueltos:

Problemas resueltos de termodinámica BARRIO CASADO, MARIA DEL, BRAVO GUIL, EDUARDO, LANA PONS, FRANCISCO JAVIER, LOPEZ PEREZ, DAVID ORENCIO, SALUD PUIG, JOSEP, TAMARIT MUR, JOSE LUIS, 2005-01-01 Trata de forma practica los aspectos fundamentales de la Termodinamica del Equilibrio al nivel correspondiente a un primer o segundo año universitario El libro se organiza en dos partes Introduccion teorico practica y Problemas multitematicos La obra incluye en total mas de 100 cuestiones y problemas completamente resueltos cuyas explicaciones se complementan con mas de 100 figuras ilustrativas **Termodinámica: 100 ejercicios y problemas** H. Lumbroso, 2021-01-10 Esta obra de Termodinamica va destinada a los estudiantes de los cursos preparatorios de las Escuelas de Ingenieros as como a los del primer ciclo de las Facultades de Ciencias Termodinámica básica. Ejercicios, 2010-03-10 Este libro pretende ser una introducción a la termodinámica básica del equilibrio y se dirige especialmente a los alumnos de primer ciclo de ciencias y enseñanzas técnicas universitarias La obra consta de un total de diez capítulos cada uno de los cuales se compone de tres partes En la primera se expone un resumen de los conceptos teóricos más relevantes en la segunda se resuelven minuciosamente un conjunto significativo de problemas con la finalidad de facilitar la consolidación de los conceptos teóricos y en la tercera se proponen una serie de problemas para que el propio alumno evalúe su grado de asimilación de los conceptos teóricos para lo cual se le proporciona la solución numérica como guía Muchos de los problemas propuestos y resueltos tienen un ámbito multitemático lo cual les confiere un mayor carácter pedagógico Con ello se pretende que el alumno adquiera una visión exhaustiva de la termodinámica básica la cual sin duda le servirá para poder acometer de forma más efectiva el tratamiento de la termodinámica más compleja *Termodinámica Técnica. Teoría y 222 ejercicios resueltos* JUÁREZ CASTELLÓ, MANUEL CELSO, MORALES ORTIZ, M^a PILAR, 2015-01-01 Este libro se dirige a los alumnos de los primeros cursos de Ingeniería que inician el estudio de la Termodinámica Sirve de fundamento para posteriores estudios en las áreas de máquinas y motores térmicos y de energía La obra se estructura en ocho capítulos El primero introduce al estudiante en los conceptos fundamentales de la Termodinámica Técnica Los capítulos segundo a quinto presentan el Primer y el Segundo Principio de la Termodinámica tanto en sistemas cerrados como en sistemas abiertos y concluyendo en ellos se presta una atención especial a los sistemas abiertos volúmenes de control pues son los más usuales en las aplicaciones de Ingeniería Por último se abordan las aplicaciones técnicas industriales el sexto capítulo contiene una introducción a los sistemas energéticos y los capítulos séptimo y octavo estudian las mezclas de gases no reactivas Psicrometría y reactivas combustión Asimismo el libro presenta un enfoque esencialmente práctico aunque ofrece pequeñas introducciones teóricas pues incluye numerosos problemas y ejercicios completamente resueltos algo difícil de encontrar en otros textos de Termodinámica Ello permite a los alumnos capacitarse en la resolución de ejercicios aprendiendo a la vez los conceptos teóricos En definitiva este texto es una excelente herramienta que permite a los alumnos adquirir las competencias necesarias para abordar con garantías estudios más

profundos en los mbitos de la energa y los motores y mquinas trmicas **Fundamentos de termodinámica técnica** Michael J. Moran, Howard N. Shapiro, 2004 Este libro ha evolucionado a lo largo de muchos aos de enseanza de la asignatura tanto para no graduados como postgraduados Explicaciones claras y completas junto a numerosos ejemplos bien desarrollados hacen el texto agradable y casi ideal para el **Problemas resueltos de fisicoqumica** Raúl Monsalvo Vázquez, Luis Alberto Pérez Monroy, 2020-05-21 Esta obra ha sido escrita como apoyo para los cursos de Fisicoqumica que se imparten en las reas de la Qumica y la Ingeniera Qumica y con este propsito presenta los conceptos fundamentales y las tcnicas de clculo bsicas de la Termodinmica la Termodinmica y la Termoqumica Con el fin de hacer de este libro una herramienta lo ms didctica posible la seleccin de los problemas incluidos y la detallada solucin presentada de cada uno de ellos fueron elaboradas a partir de la experiencia adquirida por ambos autores en la imparticin de esta materia Adems de ser resueltos con todo detalle todos los problemas presentados al principio de cada captulo se expone la teora bsica correspondiente Aprenda a aplicar la teora fundamental de la Fisicoqumica en la solucin de problemas concretos de esta materia Conozca las tcnicas de solucin ms importantes de los problemas de la Fisicoqumica Raúl Monsalvo Vázquez obtuvo el grado de Doctor en Ciencias Qumicas con Especialidades en Estadstica prctica para el Anlisis de Datos y Control Total de la Calidad en la UNAM es consultor en las reas de Gestin de Procesos y Diseo Organizacional en el sector pblico privado y social e imparte la ctedra de Procesos Industriales y de temas relacionados con la Fisicoqumica en la Unidad Profesional Interdisciplinaria de Ingeniera y Ciencias Sociales y Administrativas del IPN Luis Alberto Pérez Monroy es Ingeniero Qumico Industrial egresado del Instituto Politcnico Nacional es especialista certificado en diseo simulacin y control de reactores qumicos posee experiencia en el ramo de la industria metal mecnica e inyeccin de plstico y es catedrtico en el rea de Fisicoqumica en la Unidad Profesional Interdisciplinaria de Ingeniera y Ciencias Sociales y Administrativas del IPN

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matemático *Física para la ciencia y la tecnología* Paul A. Tipler, Gene Mosca, 2004 **College Physics** Raymond A. Serway, Jerry S. Faughn, 1999 College Physics is written for a one year course in introductory physics Preface **Física Y Química. Profesores de Enseñanza Secundaria. Temario Practico. E-book** José Vicente Morales Ortiz, 2003-10-27

Problemas de Química Física. , 2006 Esta obra que usted tiene en sus manos responde a la necesidad detectada por los autores de llegar claramente al alumno de Química Física con un manual práctico que responda a la propia naturaleza de la disciplina y le permita su correcta comprensión entendiéndola de manera precisa y completa cómo se solucionan los problemas planteados en el aula o en el laboratorio De ahí este extraordinario esfuerzo por llevar al estudiante fundamentalmente una selección de los mejores problemas de imprescindible resolución si se quiere garantizar la correcta preparación de aquel que quiera continuar y profundizar en el apasionante mundo de la Química desde las diferentes perspectivas en ciencias puras y o aplicadas Tomado de <http://bit.ly/2gZiZHU> *Física. Elementos fundamentales. Mecánica y termodinámica clásicas.*

Relatividad Mario R. Guerra, J. Correa, Ismael Nuñez, M. Scaron, 2012-01-01 Este texto consta de cinco capítulos En cada uno de ellos el lector encontrará discusiones de carácter fundamentalista de los principios de la Mecánica la Termodinámica clásica y la Relatividad especial Los capítulos tratan sobre cuestiones específicas y conocidas y aunque sus temáticas son distintas existe un denominador común a todos ellos que se extrapolan al Tomo II que consta de siete capítulos Termodinámica Cristóbal Fernández Pineda, Santiago Velasco Maíllo, 2009-09-17 No es un libro de texto al uso que responda a un programa de un determinado plan de estudios más bien con sus contenidos se pueden confeccionar e impartir eligiéndolos adecuadamente programas distintos y variados desde un nivel introductorio a un nivel más avanzado La estructura de los capítulos sigue una metodología que comprende unos contenidos principales y unos complementos En el texto principal se exponen y desarrollan las que consideramos partes básicas del capítulo mientras que en los complementos aparecen aquellas partes no de menor importancia que completan o muestran versiones alternativas al texto principal *Física. Prueba Específica. Prueba de Acceso a la Universidad Para Mayores de 25 Años.e-book* María Dolores Santana, 2003-06-04

Termodinámica técnica G. Boxer, 1979 El contenido de este libro se basa en el material recogido en un primer curso de Ingeniería mecánica de pregraduados y se presenta como una base razonablemente completa de trabajo fundamental que es esencial para cualquier estudio más avanzado de esta materia Un estudio de los sistemas termodinámicos Manuel Zamora Carranza, 1998 *Curso Sobre el Formalismo y Los Métodos de la Termodinámica* Jesús Biel Gayé, 1997 Consultar comentario general de la obra completa **Problemas resueltos de termodinámica** Javier Cañada Ribera, José Luis Gómez Ribelles, Ricardo Díaz Calleja, Ma Jesús Sanchís Sánchez, 1999

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