

2.1.4 & 2.1.5 - Acids and redox

What are standard solutions?	A solution for which its concentration is known accurately.
What 3 things should you ensure when making standard solutions?	<ol style="list-style-type: none"> 1. Ensure you have clean dry equipment. 2. Make washings to remove any salt BEFORE MAKING up the solution to the specified amount. 3. Invert the volumetric flask to ensure everything is dissolved.
How does ammonia react with acid?	<ul style="list-style-type: none"> • ammonia + acid \rightarrow ammonium salt • $NH_3(aq) + HNO_3(aq) \rightarrow NH_4NO_3(aq)$
What are strong and weak acids ? Give 3 strong acids and 1 weak acid	<ul style="list-style-type: none"> • Strong acids dissociate fully to release H^+ ions in a solution e.g. $HCl \rightarrow H^+ + Cl^-$ • Weak acids dissociate partially to release H^+ ions in a solution – exist in equilibrium • Strong acids include: HCl, H_2SO_4, and HNO_3. • Weak acids include carboxylic acids such as CH_3COOH.
What are strong and weak bases? Give 1 weak base	<ul style="list-style-type: none"> • Strong bases dissociate fully to release OH^- ions in a solution • Weak bases dissociate partially to release OH^- ions in a solution – exist in equilibrium e.g. NH_3
What are diprotic and triprotic acids (with examples) and what does this mean for neutralisation?	<ul style="list-style-type: none"> • Diprotic acids and triprotic acids donate two and three protons respectively. • Examples include: H_2SO_4 and H_3PO_4. • They need to react to the number of moles of base to be neutralised

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Mike Smith



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A Level Chemistry For OCR Pasan Witharana, 2022-06 Prepare for OCR A Level Chemistry with over 500 questions answers and mind maps covering various topics Learn everything you need to know for your exams while working on your exam technique This book aims to maximise your revision time by following the high yield revision approach Edited by a specialist OCR examiner the answers in the book follow the marking scheme answer format Learn the subject content through questions and answers Mind maps to consolidate learning for effective revision and maximise study time Practice all question styles including multiple choice extended answer and multi step calculations Learn to answer effectively to score the maximum marks Answers to all practise questions are provided Develop your exam technique and face the exam with confidence

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OCR A Level Chemistry A. Emma Poole, 2016

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Books, Richard Parsons, 2009-07-16 A2 Level Chemistry OCR A Complete Revision Practice **OCR A Level Chemistry A**

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Table of Contents Ocr Chemistry A2 Unofficial Mark Scheme 2014

1. Understanding the eBook Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - The Rise of Digital Reading Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Personalized Recommendations
 - Ocr Chemistry A2 Unofficial Mark Scheme 2014 User Reviews and Ratings
 - Ocr Chemistry A2 Unofficial Mark Scheme 2014 and Bestseller Lists
5. Accessing Ocr Chemistry A2 Unofficial Mark Scheme 2014 Free and Paid eBooks
 - Ocr Chemistry A2 Unofficial Mark Scheme 2014 Public Domain eBooks
 - Ocr Chemistry A2 Unofficial Mark Scheme 2014 eBook Subscription Services
 - Ocr Chemistry A2 Unofficial Mark Scheme 2014 Budget-Friendly Options

6. Navigating Ocr Chemistry A2 Unofficial Mark Scheme 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Ocr Chemistry A2 Unofficial Mark Scheme 2014 Compatibility with Devices
 - Ocr Chemistry A2 Unofficial Mark Scheme 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Highlighting and Note-Taking Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Interactive Elements Ocr Chemistry A2 Unofficial Mark Scheme 2014
8. Staying Engaged with Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ocr Chemistry A2 Unofficial Mark Scheme 2014
9. Balancing eBooks and Physical Books Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr Chemistry A2 Unofficial Mark Scheme 2014
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Setting Reading Goals Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Fact-Checking eBook Content of Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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