Nuclear Equations - SOLUTIONS

Bombardment of aluminum-27 by alpha particles produces phosphorous-30 and one other particle.
 Write the nuclear equation for this reaction and identify the other particle.

$$^{27}_{13}$$
 Al+ $^{4}_{2}$ $\alpha \rightarrow ^{30}_{15}$ P+ $^{1}_{0}$ n

The other particle formed is a neutron

Plutonium-239 can be produced by bombarding uranium-238 with alpha particles. How many neutrons will be produced as a by product of each reaction? Write the nuclear equation for this reaction?
 200 U+4 a → 200 Pu +3 and particles.

Three neutrons are formed for every one nucleus of U-238 bombarded.

3. When bombarded with neutrons, cobalt-59 is converted to cobalt-60. What is the nuclear equation for this reaction?

4. One method for producing plutonium-238 is by bombarding uranium-238 with deuterium (hydrogen-2), which produced neptunium-238 and 2 neutrons. The unstable neptunium then decays to form plutonium-238. Write the nuclear equations for this two-step reaction. What other particle is produced in the second reaction?

$$^{238}U + ^{2}_{1}H \rightarrow ^{238}_{33}Np + 2^{1}_{0}n$$

 $^{238}U \rightarrow ^{238}Pu + ^{9}B$

A beta particle is also produced in the second reaction.

Neutron bombardment of plutonium-239 yields americium-240 and another particle. Write the nuclear equation and identify the other particle produced.

The other particle produced is a beta particle.

Alpha-particle bombardment of plutonium-239 produced a neutron and another radionuclide. Write the nuclear equation for this reaction and identify the radionuclide.

The radionuclide formed is Cm-242.

One possible result of the impact of a neutron on a uranium-235 nucleus is the splitting of the
uranium into tellurium-137, zirconium-97 and two other particles. Write the nuclear equation for this
reaction and identify the two other particles.

$$^{235}_{92}U + ^{1}_{0}n \rightarrow ^{137}_{52}Te + ^{97}_{49}Zr + 2^{1}_{0}n$$

The two other particles formed are neutrons.

8. When bombarded with neutrons, lithium-6 produced an alpha particle and an isotope of hydrogen. Write the nuclear equation for this reaction. What isotope of hydrogen is produced?

$$^{\circ}_{3}\text{Li}+^{\circ}_{0}n\rightarrow ^{4}_{2}\alpha+^{3}_{1}\text{H}$$

The isotope formed is hydrogen-3.

With what particle would you bombard sulfur-32 to produce hydrogen-1 and phosphorus-32? Write the appropriate nuclear equation?

10. With what particle would you bombard bismuth-209 to produce astatine-211 and 2 neutrons? Express this reaction in the form of a nuclear equation.

The particle that bombards bismuth-209 is an alpha particle.

L Cohen

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- 1. Understanding the eBook Nuclear Equation For Natural Transmutations Worksheet Answers
 - The Rise of Digital Reading Nuclear Equation For Natural Transmutations Worksheet Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Nuclear Equation For Natural Transmutations Worksheet Answers
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nuclear Equation For Natural Transmutations Worksheet Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Nuclear Equation For Natural Transmutations Worksheet Answers
 - Personalized Recommendations
 - Nuclear Equation For Natural Transmutations Worksheet Answers User Reviews and Ratings

- Nuclear Equation For Natural Transmutations Worksheet Answers and Bestseller Lists
- 5. Accessing Nuclear Equation For Natural Transmutations Worksheet Answers Free and Paid eBooks
 - Nuclear Equation For Natural Transmutations Worksheet Answers Public Domain eBooks
 - Nuclear Equation For Natural Transmutations Worksheet Answers eBook Subscription Services
 - Nuclear Equation For Natural Transmutations Worksheet Answers Budget-Friendly Options
- 6. Navigating Nuclear Equation For Natural Transmutations Worksheet Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Nuclear Equation For Natural Transmutations Worksheet Answers Compatibility with Devices
 - Nuclear Equation For Natural Transmutations Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Nuclear Equation For Natural Transmutations Worksheet Answers
 - Highlighting and Note-Taking Nuclear Equation For Natural Transmutations Worksheet Answers
 - Interactive Elements Nuclear Equation For Natural Transmutations Worksheet Answers
- 8. Staying Engaged with Nuclear Equation For Natural Transmutations Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nuclear Equation For Natural Transmutations Worksheet Answers
- 9. Balancing eBooks and Physical Books Nuclear Equation For Natural Transmutations Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nuclear Equation For Natural Transmutations Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Nuclear Equation For Natural Transmutations Worksheet Answers
 - Setting Reading Goals Nuclear Equation For Natural Transmutations Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Nuclear Equation For Natural Transmutations Worksheet Answers
 - Fact-Checking eBook Content of Nuclear Equation For Natural Transmutations Worksheet Answers
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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