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Student Exploration: Solubility and Temperature

Vocabulary: concentration, dissolve, homogeneous mixture, solubility, solubility curve, solute, solution, solvent

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

- What happens when you stir a spoonful of sugar into hot water? It dissolves.
- When sugar or another substance is dissolved in water, it disappears from view and forms a homogeneous mixture with the water, also called a solution.

If you can't see the sugar, how can you tell that it is there? You wouldn't know unless you taste it since it dissolves. If the water is sweet then you'll know sugar is in it.

3. Does sugar dissolve more easily in hot water or cold water? Hot water.

Gizmo Warm-up

A solution generally consists of two parts, a **solute** that is dissolved and a **solvent** that the solute is dissolved into. For example, sugar is a solute that is dissolved into the solvent water. In the *Solubility and Temperature* GizmoTM, you will study how temperature affects now much solute will dissolve in a solution.

To begin, check that Potassium nit. "to is selected at dithe Temp, of the water is 20 °C. Clink OK.

In this solution, what is the solute? Potascium nitrate.



What is the solvent? Water

- Click Add 10 g to mix 10 g of ootass. m nitrate into the water.
 - A. Did all of the potassium in rate dissolve? Yes.
 - B. F. w can you tell? You can't see the solute (Potassium nitrate) because it's none left over

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Thomas R. Lord, Donald P. French, Linda W. Crow

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