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Atomic Structure and the Periodic Table Study Guide

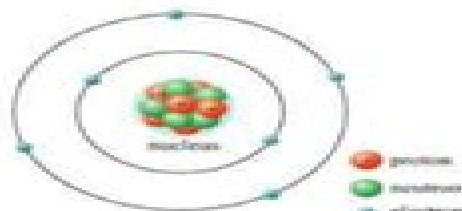
1. What is an atom?

Smallest particle of an element that retains the properties of that element.

2. Complete the following chart about subatomic particles.

Name	Charge	Symbol	Location	Relative Mass
Proton	+	p^+	Nucleus	1 amu
Electron	-	e^-	Electron cloud	0
neutron	0	n^0	nucleus	1 amu

3. Draw an atom and label the subatomic particles.

4. Which subatomic particle is used to determine the identity of an element?
protons5. Where is most of the mass of the atom? What particle(s) are in this area?
Nucleus; protons and neutrons

6. What is an atom called after it has gained or lost electrons? ___ion_____

7. What do you call an atom if it has more or less neutrons than normal? ___isotope_____

8. Label the information provided in the periodic table.

8	←	_____	Atomic number
O	←	_____	Symbol
Oxygen	←	_____	Name
15.999	←	_____	Atomic mass

9. From a box on the periodic table, how do you determine the number of protons, electrons, and neutrons?

1) # p^+ = _____ atomic number _____2) # e^- = _____ number of protons _____3) # n^0 = _____ atomic mass - atomic number _____

10. What does the atomic number represent? # protons in the nucleus

11. What does the atomic mass represent? # particles in the nucleus (protons & neutrons)

Study Guide Atoms And Elements

Tao Wei



Study Guide Atoms And Elements:

Student Notebook and Study Guide to Accompany The Human Body Bruce Wingerd, Patty Bostwick Taylor, 2013-02-01 This Student Notebook and Study Guide the ideal companion to Bruce Wingerd's The Human Body reinvents the traditional study guide by giving students a tool to help grasp information in class and reinforce learning outside of class Too often students struggle to both learn the concepts presented and simultaneously record crucial information The Student Notebook and Study Guide provides a structure for recording in class material that parallels the text's concept presentation and includes supplemental questions and activities for assignment outside of the classroom A complete answer guide for both the in class and out of class materials is available online

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