

Name: _____ Date: _____

Student Exploration: Element Builder

Background: atoms contain smaller particles, including electrons and protons. Electrons carry a small **negative** charge, while protons carry a small **positive** charge. Electrons are found in the **electron cloud** around the nucleus, while protons are found in the **nucleus**.

Directions:
After periodic table

Use the element table

- Click on neutral hydrogen atoms.
- Click on the element box in periodic table, under 1 proton, 0 neutrons, and 1 electron.



Question: After 2.2, we see atoms with more than one proton. How are these atoms different from the periodic table before this change is made?

- More elements have been added to the periodic table since.



After 2.2, the periodic table has changed the way that they are organized by number of protons and neutrons.

- More elements have been added to the periodic table since the change is made.

is more

the same

is more organized by number of protons and neutrons is more the same

is more organized by number of protons and neutrons is more the same

- More elements have been added to the periodic table since the change is made

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CH Cherryholmes



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