



Rugged Cast Iron Construction

# Power Hacksaw Manual Parts

**United States. War Department**



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web name the following molecular compounds scl<sub>2</sub> n<sub>2</sub>o<sub>4</sub> p<sub>4</sub>o<sub>10</sub> pf<sub>5</sub> scl<sub>2</sub> sulfur dichloride n<sub>2</sub>o<sub>4</sub> dinitrogen tetroxide the a of tetra is dropped before the o of oxide p<sub>4</sub>o<sub>10</sub> tetraphosphorous decoxide the a of deca is dropped before the o of oxide pf<sub>5</sub> phosphorous pentafluoride

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web one of the ions produced is always  $H^+$  which immediately combines with a water molecule to form the hydronium ion  $H_3O^+$  the  $H^+$  in is what defines the acidic properties of a substance because of this special classification acids have a naming system different from ionic or other molecular covalent compounds

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web examine the pairs of ternary acids in model 2 that contain sulfur phosphorus and nitrogen each pair has one acid that ends in ous and another that ends in ic these endings are related to the name of the polyatomic anion found in the acid ate or ite

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web the acids lose a hydrogen ion for each acid base reaction in model 2 describe the role of the bronsted lowry base in the proton  $H^+$  ion transfer that occurs the base gains or accepts a hydrogen ion proton as you saw in last week's pogil all arrhenius bases in model 1 have an  $OH^-$  ion in their chemical formulas

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web dec 10 2022  $HCl$  answer hydrochloric acid  $HBr$  answer hydrobromic acid  $H_2S$  answer hydrosulfuric acid  $HF$  answer hydrofluoric acid  $HClO_3$  answer chloric acid chlorate  $H_2SO_3$  answer sulfurous acid sulfite  $H_2SO_4$  answer sulfuric acid sulfate  $H_3PO_3$  answer phospho naming acids practice pogil answer key december 10

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