

Radiation Sources

• Problem 1.1. Radiation Energy Spectra: Line vs. Continuous

Line (or discrete energy): a, c, d, e, f, and i.

Continuous energy: b, g, and h.

• Problem 1.2. Conversion electron energies compared.

Since the electrons in outer shells are bound less tightly than those in closer shells, conversion electrons from outer shells will have greater emerging energies. Thus, the M shell electron will emerge with greater energy than a K or L shell electron.

• Problem 1.3. Nuclear decay and predicted energies.

We write the conservation of energy and momentum equations and solve them for the energy of the alpha particle. Momentum is given the symbol "p", and energy is "E". For the subscripts, "a" stands for alpha, while "b" denotes the daughter nucleus.

$$p_a + p_b = 0 \quad \frac{p_a^2}{2m_a} = E_a \quad \frac{p_b^2}{2m_b} = E_b \quad E_a + E_b = Q \quad \text{and} \quad Q = 5.5 \text{ MeV}$$

Solving our system of equations for E_a , E_b , p_a , p_b , we get the solutions shown below. Note that we have two possible sets of solutions (this does not effect the final result).

$$E_b = 5.5 \left(1 - \frac{m_a}{m_a + m_b} \right) \quad E_a = \frac{5.5 m_a}{m_a + m_b}$$

$$p_a = \pm \frac{3.31662 \sqrt{m_a} \sqrt{m_b}}{\sqrt{m_a + m_b}} \quad p_b = \pm \frac{3.31662 \sqrt{m_a} \sqrt{m_b}}{\sqrt{m_a + m_b}}$$

We are interested in finding the energy of the alpha particle in this problem, and since we know the mass of the alpha particle and the daughter nucleus, the result is easily found. By substituting our known values of $m_a = 4$ and $m_b = 206$ into our derived E_a equation we get:

$$E_a = 5.395 \text{ MeV}$$

Note : We can obtain solutions for all the variables by substituting $m_b = 206$ and $m_a = 4$ into the derived equations above :

$$E_a = 5.395 \text{ MeV} \quad E_b = 0.105 \text{ MeV} \quad p_a = \pm 6.570 \sqrt{amu + \text{MeV}} \quad p_b = \pm 6.570 \sqrt{amu + \text{MeV}}$$

• Problem 1.4. Calculation of Wavelength from Energy.

Since an x-ray must essentially be created by the de-excitation of a single electron, the maximum energy of an x-ray emitted in a tube operating at a potential of 195 kV must be 195 keV. Therefore, we can use the equation $E=h\nu$, which is also $E=hc/\lambda$, or $\lambda=hc/E$. Plugging in our maximum energy value into this equation gives the minimum x-ray wavelength.

$$\lambda = \frac{h \times c}{E} \quad \text{where we substitute } h = 6.626 \times 10^{-34} \text{ J} \cdot \text{s}, \quad c = 299\,792\,458 \text{ m/s and } E = 195 \text{ keV}$$

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