

# Solutions to the Exercices of

## *C. J. Foot's Atomic Physics*

Chenchao Zhao

*Department of Physics, Beijing Normal University, Beijing, China*

(Dated: June 28, 2011)

### 1 Early atomic physics

Key formulas:

$$\frac{1}{\lambda} = R \left( \frac{1}{n^2} - \frac{1}{n'^2} \right) \quad (1.1)$$

$$R_H = R \frac{m_H}{m_e + m_H} \quad (1.2)$$

$$a_0 = \frac{\hbar^2}{(Ze^2/4\pi\epsilon_0)m_e} = 0.529 \times 10^{-10} \text{ m} \quad (1.3)$$

$$E = - \frac{Ze^2/4\pi\epsilon_0}{2a_0} \frac{1}{n^2} \propto Z^2 \quad (1.4)$$

$$\Delta E = \frac{\alpha^2}{n^2} E \quad (1.5)$$

$$\alpha = \frac{e^2/4\pi\epsilon_0}{\hbar c} = 1/137 \quad (1.6)$$

$$\frac{f}{c} = R_\infty ((Z - \sigma_K)^2 - (Z - \sigma_L)^2) \quad (1.7)$$

$$\rho(\omega) = \frac{\hbar\omega^3}{\pi^2 c^3} \frac{1}{\exp \hbar\omega/kT - 1} \quad (1.8)$$

$$\frac{N_2}{g_2} = \frac{N_1}{g_1} \exp \left( - \frac{\hbar\omega}{kT} \right) \quad (1.9)$$

$$P = \frac{e^2 E_0^2 \omega^4}{12\pi\epsilon_0 c^3} \quad (1.10)$$

$$\Omega_L = \frac{eB}{2m_e} \quad (1.11)$$

$$\tau = \frac{6\pi\epsilon_0 m_e c^3}{e^2 \omega^3} \quad (1.12)$$

# Solution Manual Of Atomic Physics Foot

**Marcel A. Müller**



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## **Table of Contents Solution Manual Of Atomic Physics Foot**

1. Understanding the eBook Solution Manual Of Atomic Physics Foot
  - The Rise of Digital Reading Solution Manual Of Atomic Physics Foot
  - Advantages of eBooks Over Traditional Books
2. Identifying Solution Manual Of Atomic Physics Foot
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Solution Manual Of Atomic Physics Foot
  - User-Friendly Interface
4. Exploring eBook Recommendations from Solution Manual Of Atomic Physics Foot
  - Personalized Recommendations
  - Solution Manual Of Atomic Physics Foot User Reviews and Ratings
  - Solution Manual Of Atomic Physics Foot and Bestseller Lists
5. Accessing Solution Manual Of Atomic Physics Foot Free and Paid eBooks

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    - Solution Manual Of Atomic Physics Foot Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Solution Manual Of Atomic Physics Foot
    - Highlighting and Note-Taking Solution Manual Of Atomic Physics Foot
    - Interactive Elements Solution Manual Of Atomic Physics Foot
  8. Staying Engaged with Solution Manual Of Atomic Physics Foot
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Solution Manual Of Atomic Physics Foot
  9. Balancing eBooks and Physical Books Solution Manual Of Atomic Physics Foot
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Solution Manual Of Atomic Physics Foot
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Solution Manual Of Atomic Physics Foot
    - Setting Reading Goals Solution Manual Of Atomic Physics Foot
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Solution Manual Of Atomic Physics Foot
    - Fact-Checking eBook Content of Solution Manual Of Atomic Physics Foot
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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