Kf: formation constant
- equilibrium constant for the formation of the complex ion.
question: Ag+ 2 NHz = Ag (NHz) = EF= 1.7.107

question: Ag+ 2NH3 = Ag (NH3)2 EF= 1.7.107
- calculater AgNH32-Ag ration at 0.10NH3
1.7.107 = (Ag(NH3)2)7
TNH.74(Aa7)

1.7.10° (NH3)2] [Ag] 1.7.10° (0.1)2 - 2 1.7.10° = Ag(NH3)2 Ag

question#2: calculate equilibrium constant:

Ag $(NH_3)_2 + 75_2O_7 \approx Ag(S_2O_7)_2 + 2NH_3$ Ag $(NH_3)_2 \rightarrow Ag + 2NH_3 - 4000$ Ag $+2S_2O_3 \rightarrow Ag(S_2O_5) - 1.40013$

K= 1.10" . 1.7.10"

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