

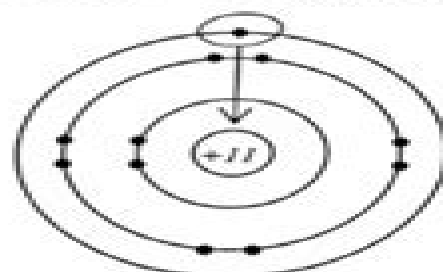
Advanced Periodic Trends

How do the particles in an atom interact to dictate periodic properties of elements?

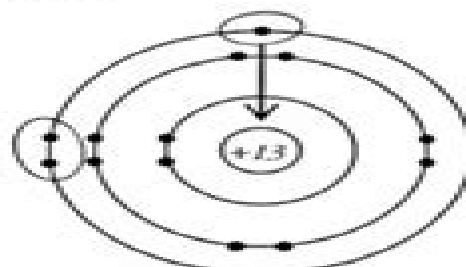
Why?

Previously you have learned about Coulombic attraction and how it governs the trends in properties among the elements of the Periodic Table. For example, as you move across a period, atoms have more protons in the nucleus, which pulls the electrons in tighter making smaller, more electronegative atoms. As you proceed down a column of the periodic table, the valence electrons get further away, and the atoms get larger and less electronegative. Inquisitive students will always ask, "If more protons in the nucleus pull electrons in tighter, then why don't atoms get smaller when going down a column?" That is a good question, and one that you will investigate in this activity.

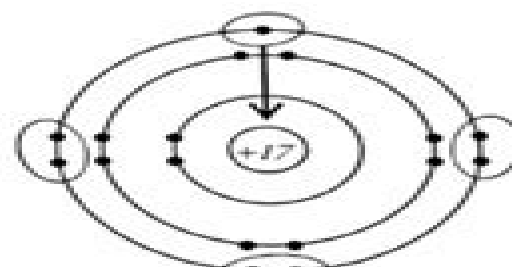
Model 1 – Period 3 Elements



Sodium



Aluminum



Chlorine

Atomic Number	11	13	17
Number of Electrons	11	13	17
Number of Valence Electrons	1	3	7
Core Charge	+1	+3	+7
Atomic Radius	186 pm	143 pm	99 pm
1st Ionization Energy	496 kJ/mol	578 kJ/mol	1251 kJ/mol
Electronegativity	0.9	1.5	3.0

1. Describe the relationship of the three elements in Model 1 with regard to their relative positions on the periodic table.

The three elements are in the same row of the periodic table.

2. Refer to a periodic table to complete the first three rows of the table in Model 1.

See Model 1.

Pogil Periodic Trends

Kit Chapman

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located on the right side of the slide.

Pogil Periodic Trends:

Chemistry Education Javier García-Martínez, Elena Serrano-Torregrosa, 2015-05-04 Winner of the CHOICE Outstanding Academic Title 2017 Award This comprehensive collection of top level contributions provides a thorough review of the vibrant field of chemistry education Highly experienced chemistry professors and education experts cover the latest developments in chemistry learning and teaching as well as the pivotal role of chemistry for shaping a more sustainable future Adopting a practice oriented approach the current challenges and opportunities posed by chemistry education are critically discussed highlighting the pitfalls that can occur in teaching chemistry and how to circumvent them The main topics discussed include best practices project based education blended learning and the role of technology including e learning and science visualization Hands on recommendations on how to optimally implement innovative strategies of teaching chemistry at university and high school levels make this book an essential resource for anybody interested in either teaching or learning chemistry more effectively from experience chemistry professors to secondary school teachers from educators with no formal training in didactics to frustrated chemistry students

Chemistry and Our Universe, Return to the periodic table introduced in Lecture 1 to practice predicting properties of elements based on their electronic structure Then witness what happens when three different alkali metals react with water Theory forecasts a pronounced difference in the result Is there

Chemical Periodicity Robert Thomas Sanderson, 2013-04-20 *Mastering the Periodic Table* Cybellium Ltd, 2024-10-26 Designed for professionals students and enthusiasts alike our comprehensive books empower you to stay ahead in a rapidly evolving digital world Expert Insights Our books provide deep actionable insights that bridge the gap between theory and practical application Up to Date Content Stay current with the latest advancements trends and best practices in IT AI Cybersecurity Business Economics and Science Each guide is regularly updated to reflect the newest developments and challenges Comprehensive Coverage Whether you re a beginner or an advanced learner Cybellium books cover a wide range of topics from foundational principles to specialized knowledge tailored to your level of expertise Become part of a global network of learners and professionals who trust Cybellium to guide their educational journey www.cybellium.com

Periodic Table 91 Success Secrets - 91 Most Asked Questions on Periodic Table - What You Need to Know Patrick Stein, 2014-10-01 Probably The Best Periodic table Guide To Date There has never been a Periodic table Guide like this It contains 91 answers much more than you can imagine comprehensive answers and extensive details and references with insights that have never before been offered in print Get the information you need fast This all embracing guide offers a thorough view of key knowledge and detailed insight This Guide introduces what you want to know about Periodic table A quick look inside of some of the subjects covered Extended periodic table Predicted properties of undiscovered elements Extended periodic table Elements 165 to 172 Extended periodic table Eka superactinides Extended periodic table 8s elements Transition metal Position in the Periodic Table Block periodic table Relativistic effect Periodic table deviations

History of the periodic table Ancient times Periodic table Period 1 Chemical element The periodic table Periodic table Period 8 Periodic table Blocks Periodic table Further development Group periodic table Periodic table Ionization energy Extended periodic table large version Periodic table Period 3 Fricke model Extended periodic table Alternative periodic tables Elements repeating Ronald L Rich 2005 Periodic table Other conventions and variations Wide periodic table large version Periodic table electron configurations Extended periodic table 7d transition metals Neutronium Neutronium and the periodic table The Periodic Table of Videos Development Periodic table Layout History of the periodic table Alexandre Emile Beguyer de Chancourtois Mendeleev Periodic table Alternative periodic tables Other Periodic table crystal structure Body centred cubic Periodic table Electron configuration and much more [The Periodic Table](#) D. H. Rouvray, R. Bruce King, 2004 This book starts with chapters that trace the early history and development of the Periodic Table The subsequent development of the Table is then presented in chapters that discuss the structure and characteristics of the Table probe its group theoretical and quantum theoretical basis examine its foundations and explore its many uses and applications Midwest **Periodic Table in Minutes** Dan Green, 2016-05-19 An icon of science the Periodic Table defines the fundamental chemistry of everything in the universe In this compact yet comprehensive guide Dan Green outlines the history development and workings of the table shows how its design reflects and illuminates the organisation of all matter and even explains what it has to tell us about the chemistry of distant stars and of our own bodies Contents include an individual entry for every known element detailing properties uses and key data and sections on the patterns and groups of the famous table as well as explanations of basic chemistry concepts such as elements and compounds atomic structure chemical bonds reactions and radioactivity amongst many others **Trends in the Periodic Table** Open University. S25-Course Team, 1972 *The Mathematics of the Periodic Table* D. H. Rouvray, R. Bruce King, 2006 The Periodic Table effectively embraces the whole realm of chemistry within the confines of one comparatively simple and easily understood chart of the chemical elements Over many years the Periodic Table has proven to be indispensable not only to chemists of all kinds but also to a host of other scientists including biologists geologists and physicists It is thus hardly surprising that the Periodic Table has become one of our most celebrated contemporary scientific icons In the present work various aspects of the Periodic Table that are seldom if ever featured elsewhere are given prominence The twelve presentations contained herein all have a mathematical flavour because it is the intention to highlight the often neglected mathematical features of the Periodic Table and several closely related topics The book starts out by considering predictions of what the ultimate size of the Periodic Table will be when all of the possible artificial chemical elements have been synthesised It then moves on to an examination of the nature of the periodicity extant in the Periodic Table and some methods for the prediction of the properties of the super heavy elements The Periodic Table is next explored in various dimensions other than two The natural clustering of the elements into groups is studied by three different but complementary routes namely via the topological

structures of the groups the self association of the elements as evidenced by neural network studies and information theoretical analysis of the behaviour of atoms Following a detailed investigation of the mathematical basis for the periodicity seen in atomic and molecular spectroscopy three separate presentations delve into many different aspects of the group theoretical structure of the Periodic Table The unusual combination of themes offered here will appeal to all who seek a more detailed and intimate knowledge of the Periodic Table than that available in standard texts on the subject

Trends in the Periodic Table Open University. Structure, Bonding and the Periodic Law Course Team,1973

Periodic Tables Unifying Living Organisms At The Molecular Level: The Predictive Power Of The Law Of Periodicity Antonio Lima-de-faria,2017-11-08

The DNA sequencing of a series of living organisms has elucidated many biological problems But the internal atomic and electronic evolution of DNA remains to be mapped in detail RNA and DNA now appear to be the prime determinants of biological evolution leading to the sudden appearance of novel organism structures and functions that emerge ready made as a surprise to the organism This has been demonstrated by the manipulation of genes that led to the sudden production of additional complete wings and legs in flies and birds The study of this internal atomic construction of macromolecules is being investigated at the large electron accelerators such as the MAX IV Synchrotron Radiation Laboratory Lund University Sweden The periodicity of the chemical elements is well known from its iconic Table Significantly this periodicity can now be seen to extend to the properties of living organisms Biological properties as different as flight vision luminescence and regeneration as well as others show unexpectedly periodic emergence They resurface without previous announcement in most unrelated plant and animal families and they emerge irrespective of whether the organism is a simple invertebrate or a most complex mammal Moreover this periodicity does not necessarily start at the cell or DNA levels but appears initially in crystals and minerals where it is shown to be a pure atomic and electronic process e g in luminescence and regeneration The assembled molecular evidence led to the construction of Periodic Tables of living organisms placing them in a position comparable to the periodicity of the chemical elements Surprisingly there are striking resemblances between the periodicities of the chemical elements and those of living organisms In addition the two types of Tables increase our insight into the events directing atomic evolution since the periodic law established in chemical elements turns out to be applicable to the periodicity of living organisms The new Periodic Tables introduce a predictive capacity in biological evolution that before was hardly contemplated Eric Scerri from the Department of Chemistry and Biochemistry California University Los Angeles who is the Author of the book *The Periodic Table Its Story and its Significance* Oxford University Press stated in an e mail that Professor Lima de Faria s book is wonderful and a pioneering work

Table Manners for Beginners Jim Ross,2013-07 The Ross Lattner IntuitivChemistry series is different The Ross Periodic Table is a powerful new way to approach the periodic table aimed at high school students in their first chemistry course How does it work Every atom of the main block elements is represented with just three features the atomic core the valence radius and the valence electrons

Using only their natural intuitive reasoning students make use of these three features to make accurate predictions of chemical behavior Students will quickly learn these three simple concepts and apply them to an astonishing number of chemical phenomena Within a few days they will accurately predict and explain the behavior of the elements covered in introductory chemistry courses These concepts will form a strong foundation for all the chemistry you ll teach in high school Your students will gain a level of fluency that is not possible using other approaches As teachers and parents know personal competency is the only reliable foundation for building self esteem and confidence Compatible with the emerging consensus on the nature of science Works with not against students natural cognitive growth Integrates language with learning

Facilitating Conceptual Change in Students' Understanding of the Periodic Table Mansoor Niaz, Marniev

Luiggi, 2013-07-25 This book is about how students are taught the periodic table It reviews aspects of the periodic table s development using the history and philosophy of science The teaching method presented in this book is ideal for teaching the subject in high school and at introductory university level Chemistry students taught in this new experimental way are compared with those taught in the traditional way and the author describes how tests found more conceptual responses from the experimental group than the control group The historical aspects of importance to this teaching method are the role of the Karlsruhe Congress of 1860 the accommodation of the chemical elements in the periodic table prediction of elements that were discovered later corrections of atomic weights periodicity in the periodic table as a function of the atomic theory and the accommodation of argon The experimental group of students participated in various activities including discussion of various aspects related to the history and philosophy of science construction of concept maps and their evaluation by the students PowerPoint presentations and interviews with volunteer students

The Knowledge: The Periodic Table Paul Strathern, 2015-06-18 Introducing a new series of information led pocket guides aimed at demystifying a wide and eclectic range of subjects Called The knowledge each book within the series is written by an expert in the field providing a lively and informative introduction to their respective topic The formulation of the periodic table in 1869 revolutionised chemistry in the same way that Darwin and Newton s theories had advanced the fields of biology and physics The discovery of a relationship between the known elements revealed a link between the scientific disciplines and offered scientists an inkling into the blueprint of the universe This fascinating book traces the story of the table of elements

Periodic Table And Periodic Properties Arora M G, M. G. Arora, 1997-01-01 Periodic Table Is The Essence Basis The Systematic And Scientific Study Of Chemistry Physics And Even Biological Sciences Though Aplenty Of Literature On The Subject Is Available Scattered Here And There The Present Book Is Unique Which Discusses Periodic Table And Periodic Properties Elaborately Students Of Undergraduate And Postgraduate Classes Researchers And Teachers Of Chemistry And Physics Will Find This Book Most Useful And Informative

Superheavy Kit Chapman, 2019-06-13 SHORTLISTED FOR THE 2020 AAAS SUBARU SB F PRIZE FOR EXCELLENCE IN SCIENCE BOOKS How new elements are discovered why they matter and where they will

take us Creating an element is no easy feat It s the equivalent of firing six trillion bullets a second at a needle in a haystack hoping the bullet and needle somehow fuse together then catching it in less than a thousandth of a second after which it s gone forever Welcome to the world of the superheavy elements a realm where scientists use giant machines and spend years trying to make a single atom of mysterious artefacts that have never existed on Earth From the first elements past uranium and their role in the atomic bomb to the latest discoveries stretching the bounds of our chemical world Superheavy reveals the hidden stories lurking at the edges of the periodic table Why did US Air Force fly planes into mushroom clouds Who won the transfermium wars How did an earthquake help give Japan its first element And what happened when Superman almost spilled nuclear secrets In a globe trotting adventure that stretches from the United States to Russia Sweden to Australia Superheavy is your guide to the amazing science filling in the missing pieces of the periodic table You ll not only marvel at how nuclear science has changed our lives you ll wonder where it s going to take us in the future **Periodic Table**

Advanced BarCharts, Inc,2014 **Chemistry: The Periodic Table and Periodicity** ,2015 Animation succinctly develops the relationship between the electronic structure of an atom and its properties demonstrating clearly why there are families of elements and gradual changes in the properties of elements arranged by atomic number across the table **Periodicity and the S- and P- Block Elements** Nicholas C. Norman,2021 The renowned Oxford Chemistry Primers series which provides focused introductions to a range of important topics in chemistry has been refreshed and updated to suit the needs of today s students lecturers and postgraduate researchers The rigorous yet accessible treatment of each subject area is ideal for those wanting a primer in a given topic to prepare them for more advanced study or research Moreover cutting edge examples and applications throughout the texts show the relevance of the chemistry being described to current research and industry The learning features provided including end of chapter questions and online multiple choice questions encourage active learning and promote understanding Furthermore frequent diagrams margin notes further reading and glossary definitions all help to enhance a student s understanding of these essential areas of chemistry This new and updated edition of Periodicity and the s and p Block Elements provides a compelling and accessible introduction to key periodic trends found within the s and p blocks of the periodic table and includes coverage of the elements themselves as well as the compounds they form Additional chapters focus of acidity and basicity as well as on structure The final chapter is entirely new to the second edition and contains a critical examination of many theories models and approaches to the study of the ideas explored in the book Digital formats and resourcesThe second edition is available for students and institutions to purchase in a variety of formats and is supported by online resources The e book offers a mobile experience and convenient access along with functionality tools navigation features and links that offer extra learning support www.oxfordtextbooks.co.uk/ebooks Online resources include multiple choice questions for students to check their understanding and for registered adopters figures and tables from the book **How to Use the Periodic Table to Find Hidden Cycles** ,1990

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