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# Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

There are eight markers in a full set, but Flora and Frank each only have seven markers.
 Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? \_\_\_\_Share\_\_\_\_

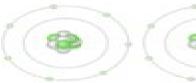
Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

share 2

#### Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the Covalent Bonds Gizmo, atoms form bonds in this way.

To begin, check that Fluorine is selected from the Select a substance menu. Click Play ( ) to see the electrons orbiting the nucleus of each atom.





- The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have?
- Click Pause (III). Drag an electron from the left atom to the right atom. Click Play.

What happens? it orbits both atoms

3. Click Pause, drag an electron from the right atom to the left, and then click Play.

What happens now? it orbits both atoms

# **Student Exploration Covalent Bonds Worksheet**

**Daniela Niemeyer** 

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#### **Table of Contents Student Exploration Covalent Bonds Worksheet**

- 1. Understanding the eBook Student Exploration Covalent Bonds Worksheet
  - The Rise of Digital Reading Student Exploration Covalent Bonds Worksheet
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Student Exploration Covalent Bonds Worksheet
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Student Exploration Covalent Bonds Worksheet
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Student Exploration Covalent Bonds Worksheet
  - Personalized Recommendations
  - Student Exploration Covalent Bonds Worksheet User Reviews and Ratings
  - Student Exploration Covalent Bonds Worksheet and Bestseller Lists
- 5. Accessing Student Exploration Covalent Bonds Worksheet Free and Paid eBooks
  - Student Exploration Covalent Bonds Worksheet Public Domain eBooks
  - Student Exploration Covalent Bonds Worksheet eBook Subscription Services
  - Student Exploration Covalent Bonds Worksheet Budget-Friendly Options

- 6. Navigating Student Exploration Covalent Bonds Worksheet eBook Formats
  - o ePub, PDF, MOBI, and More
  - Student Exploration Covalent Bonds Worksheet Compatibility with Devices
  - Student Exploration Covalent Bonds Worksheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Student Exploration Covalent Bonds Worksheet
  - Highlighting and Note-Taking Student Exploration Covalent Bonds Worksheet
  - Interactive Elements Student Exploration Covalent Bonds Worksheet
- 8. Staying Engaged with Student Exploration Covalent Bonds Worksheet
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Student Exploration Covalent Bonds Worksheet
- 9. Balancing eBooks and Physical Books Student Exploration Covalent Bonds Worksheet
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Student Exploration Covalent Bonds Worksheet
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Student Exploration Covalent Bonds Worksheet
  - Setting Reading Goals Student Exploration Covalent Bonds Worksheet
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Student Exploration Covalent Bonds Worksheet
  - Fact-Checking eBook Content of Student Exploration Covalent Bonds Worksheet
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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