

Lesson Quiz

Observing Chemical Change

Write the letter of the correct answer on the line at the left.

1. ____ Which of the following is true about chemical reactions?
A They are accompanied by changes in energy.
B They form new substances with new properties.
C both A and B
D neither A nor B
2. ____ In an endothermic reaction, energy is
A absorbed
B released
C converted to mass
D synthesized
3. ____ Which of the following is NOT a physical property?
A melting point
B state of matter
C density
D flammability
4. ____ Substances formed as a result of a chemical reaction are called
A catalysts
B precipitates
C products
D reactants

If the statement is true, write *true*. If the statement is false, change the underlined word or words to make the statement true.

5. _____ In an exothermic reaction, products have more energy than reactants.
6. _____ Water boils at 100°C. This is an example of a chemical property.
7. _____ Substances that enter into a chemical reaction are called products.
8. _____ The ability to react with oxygen is an example of a chemical property.
9. _____ Another name for a chemical change is a chemical bond.
10. _____ In a physical change, some of the physical properties of the substance may be altered and the chemical composition remains the same.

Observing Chemical Change Guided Reading And Study Guide Answers

Michal Rosen-Zvi



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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Prentice Hall Science Explorer: Teacher's ed**, 2005

Prentice Hall Science Explorer Michael J. Padilla, 2002 *Scientific American*, 1896 **The Software Encyclopedia**, 1986 *Chemical Principles in Practice* Jerry A. Bell, 1967 This forward looking laboratory text includes thirty three experiments designed to serve high level introductory chemistry courses It is intended in part to give the students practice in techniques not only those generally considered the techniques of chemistry but also those of mathematics However the only prerequisite is a good background in high school mathematics Ideally suited for use with University Chemistry or College Chemistry by Bruce Mahan this book follows the physical chemistry approach of these and most other modern chemistry texts maintaining the same flavor emphasis and degree of sophistication All experiments are open ended and include suggestions for following them up There is considerable emphasis on the involvement of the student in planning his own procedures and quantitative thinking Publisher **Resources in Education**, 1996 **Social Psychology** Graham M Vaughan, Michael A Hogg, 2013-10-16 The 7th edition of this best selling social psychology text by Graham Vaughan and Michael Hogg Social Psychology retains the structure and approach of the previous edition but has been revised to reflect the changes in the field with the material thoroughly updated throughout Social Psychology 7e continues to capture the scope and detail of contemporary social psychology as an international scientific enterprise and at the same time deals with the subject in a way that is relevant to university teaching and social psychology research in Australia and New Zealand

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