

OCR AS Chemistry

3.1.2 Group 2

Topic 23 – Group 2



Includes

- Complete lessons
- Animated PowerPoint presentation (46 slides)
- Examples with clear visual explanations
- Worksheet (11 pages)
- Answers to all questions

**Ideal for
classroom or
home learning**

Animated and interactive slides

Reaction of calcium with water

	Ca	+	2 H ₂ O	→	Ca(OH) ₂	+	H ₂	
Ca	0				+2			oxidised or reduced? oxidised 0 → +2
H			4 × (+1)		2 × (+1)		2 × 0	reduced +1 → 0
O			2 × (-2)		2 × (-2)			no change
Total	0		0		0		0	redox reaction? yes

Ca has been oxidised and **H** has been reduced.
O has not changed oxidation number.

Solubility of group 2 hydroxides

Group 2 hydroxides dissolve in water to form alkaline solutions. The more soluble the hydroxide, the more concentrated the OH⁻ ions in the resulting solution and the more alkaline it will be (higher pH).

Trend down group 2 is solubility of hydroxides

Mg(OH)₂
↓ solubility increases
Ca(OH)₂
↓ solubility increases
Sr(OH)₂
↓ solubility increases
Ba(OH)₂

Note: pH and the pH of the resulting solution increases down group 2.

Reaction of calcium oxide with water



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Rob Ritchie,Dave Gent



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