

# KEY

Name: \_\_\_\_\_

Period: \_\_\_\_\_ Date: \_\_\_\_\_

## Physics: Unit 7 Test Review

Answer the following showing ALL work including equations used, substitutions with units, and correct answer. For short answer questions, just fill in the blank.

Vocabulary: Be able to define and/or explain the following terms.

- |                   |                                  |
|-------------------|----------------------------------|
| 1. Thermal Energy | 9. Conservation of Energy        |
| 2. Conduction     | 10. Boiling/Vaporization         |
| 3. Convection     | 11. Latent Heat of vaporization  |
| 4. Radiation      | 12. Melting                      |
| 5. Temperature    | 13. Latent Heat of Fusion        |
| 6. Heat           | 14. First Law of Thermodynamics  |
| 7. Specific heat  | 15. Second Law of Thermodynamics |
| 8. Calorimeter    | 16. Entropy                      |

Multiple Choice:

- Which of the following is proportional to the kinetic energy of atoms and molecules?  
a. elastic energy      **b. temperature**      c. potential energy      d. thermal equilibrium
- Heat flow occurs between two bodies in thermal contact when they differ in which of the following properties?  
a. mass      b. density      c. specific heat      **d. temperature**
- If two small beakers of water, one at 20 °C and one at 30 °C, are emptied into a large beaker, what is the final temperature of the water?  
a. less than 20 °C  
**b. between 20 °C and 30 °C**  
c. greater than 30 °C  
d. The water temperature will fluctuate.
- Jacob takes six ice cubes from the freezer and puts four of them into a glass of water. He leaves two on the counter-top. He stirs the water until the ice cubes are much smaller and have just stopped melting. What is the most likely temperature of the water at this stage?  
a. -10 °C      **b. 0 °C**      c. 5 °C      d. 10 °C
- The ice cubes Jacob left on the counter are now almost completely melted and lying in a puddle of water. What is the most likely temperature of these smaller ice cubes?  
a. -10 °C      **b. 0 °C**      c. 5 °C      d. 10 °C
- On the stove is a kettle full of water. The water has started to boil rapidly. The most likely temperature of the water is about  
a. 88 °C      **b. 100 °C**      c. 110 °C      d. None of the above
- Five minutes later, the water in the kettle is still boiling. The most likely temperature of the water is about  
a. 88 °C      **b. 100 °C**      c. 110 °C      d. 120 °C
- What is the likely temperature of the steam above the boiling water in the kettle?  
a. 88 °C      **b. 100 °C**      c. 110 °C      d. 120 °C
- Erika takes two cups of water at 40 °C and mixes them with one cup of water at 10 °C. What is the most likely temperature of the mixture?  
a. 20 °C      b. 25 °C      **c. 30 °C**      d. 50 °C
- Gracie takes a can of coke and plastic bottle of coke from the refrigerator, where they have both been overnight. She quickly puts a thermometer in the can. The temperature is 7 °C. What are the most likely temperatures of the plastic bottle and its contents?  
a. They are both less than 7 °C  
**b. They are both at 7 °C**  
c. They are both greater than 7 °C  
d. It depends on the size of the bottle

# Thermal Energy And Thermodynamics Test Review Answers

**Wenbin Ji**



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