

Quiz & Worksheet - Drawing & Labeling Enthalpy Diagrams

1.

Which of the following enthalpy diagrams best depicts the following reaction:

Glucose + Oxygen goes to carbon dioxide and water. $\Delta H = -298 \text{ kJ/mol}$ and activation energy = 70 kJ/mol

- ☐ 
- ☐ 
- ☐ 
- ☐ 

2. What is the activation energy on an enthalpy diagram?

- ☐ The first curve up, the energy needed to start the reaction
- ☐ The first point on the diagram, the energy needed to start the reaction
- ☐ The difference between the starting and ending materials, the total change in energy
- ☐ The last point on the diagram, the change in energy

3. What is the best way to describe what's going on during a chemical reaction?

- ☐ A change of energy
- ☐ A change of matter
- ☐ A change in the fundamental properties of life
- ☐ Nothing

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Thermochemistry Energy Diagrams Worksheet Test Review Answers

S Ben Porath



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