

A chromatographic paper was folded in to a cylinder and the ends attached with small paper pins. The paper cylinder was then placed into the developing chamber. The plastic wrap was removed immediately the solvent front moved no within 1.7 cm of the top of the chromatographic paper. The $R_{f, \text{max}}$ was measured and recorded on the report sheet. Immediately the paper was removed from developing chamber, the position of the solvent front was marked.

The paper and the circle was analyzed with pencil to record any colored bands for both known and unknown solutions. The color of each metal ion was recorded on the report sheet. The chromatogram was turned to the front level so as to observe the appearance and the location of the bands. The color of each metal ion in the unknown chamber was recorded. The capillary tubes were used to spot the center of each band. The distance between the origin and the spot of each metal ion was measured and recorded. Finally, the R_f value of each metal ion was calculated and recorded.

Results and Calculations

Analysis of chromatogram of individual ions

	Color of Original	Color of $R_{f, \text{max}}$	Color of R_f	$R_{f, \text{max}}$	R_f
Cu^{2+}	Blue	Blue		0.45	0.45
Fe^{3+}	Yellow	Yellow		0.45	0.45
Ni^{2+}	Green	Green		0.45	0.45
Zn^{2+}	White	White		0.45	0.45

Conclusion

Paper And Ion Exchange Chromatography Lab Report

Michael Seilmaier



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