

<b>Skills</b> Short part: Discuss in detail / In-depth discussion / Describe / Critically discuss / Analyse / Compare / Distinguish / Evaluate / Differentiate / Explain (Discuss / Debate)	Max 20
<b>Assessment:</b> Use your answer / Critically discuss / Evaluate / Critically evaluate / Draw a graph and explain / Use the graph given and explain / Complete the given graph / Calculate / Debate / Compare / Explain / Distinguish / Interpret / Briefly discuss / <b>Summarise</b>	Max 20
<b>Guidance</b> Any relevant higher-order competences that should include: <b>Substantives</b> of what has been discussed without repeating facts already mentioned in the facts Any opinion or value judgement on the facts discussed Additional support information to strengthen the discussion / analysis A concluding statement with evaluation, if so required	Max 2
<b>Total:</b>	<b>40 Marks</b>

**QUESTION 1: MACROECONOMICS**

- Discuss in detail the phases of a business cycle with the aid of a diagram. (24 marks)
- Evaluate the expansionary reasons for changes in the business cycle. (26 marks)

**QUESTION 2: ECONOMIC PURSUITS**

- Fully discuss the main factors that impact on South African labour force growth and composition. (24 Marks)
- Explain how surplus production, trade and transport development is used in the evolution of markets. (26 marks)

**TOTAL SECTION C: [40]**

**\*\*\* TOTAL: 200 MARKS \*\*\***

**Grade 10 Economics**  
**November 2022 – Paper 1**  
**100% Editable**

# Paper 1 Economics Grade 10 2014 Final Exam

**Judd E. Hollander**



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web this experiment was being performed so that dibenzalacetone could be synthesized from benzaldehyde and acetone this experiment was performed to show how a ketone and an aldehyde could be added together through the aldol condensation the aldol condensation is extremely important because it can form a  $\beta$  hydroxy aldehyde or

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web an aldol reaction was performed by combining acetone and benzaldehyde to make dibenzalacetone the percent yield of dibenzalacetone was found to be 73 which is a good percent yield the melting point of dibenzalacetone was around the range of 104 110 c this helped conclude that dibenzalacetone had a trans trans e e isomer

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