Report Sheet - Lab 10

Date		
Section	Name	
Instructor	Team	

A. Polarity of Solutes and Solvents

Solute	Soluble/Not Soluble in	Polar or Nonpolar (A.3)	
KMnO _e		Cyclohexane (A.2)	Polar
l ₂	Soluble Non soluble	Non soluble	Non Polar
Sucrose		Non soluble	Polar
Vegetable oil	Nonsoluble	Soluble	Non Polar

Questions and Problems

Q.1 NaCl is soluble in water, but I2 is not. Explain.

Nacl is soluble in water because the solvent & solute are both polar. Iz is a non polar solute.

B. Solubility of KNO₃

B.) Mass of Container	B.2 Mass of Container + KNO,	Mass of KNO ₂	B.3 Temperature (Crystals Appear)	B.4 Solubility (g KNO ₃ /100 mL H ₂ O)

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