

STOICHIOMETRY

This section explains how to calculate the masses of neutrons required or produced, based on a stoichiometric process. It teaches you how to interpret chemical equations in terms of interacting nuclei, representative particles, moles, and mass values of NTP.

1. How can you determine the quantities of reactants and products in a chemical reaction?

^aValues are means \pm SD. ^bValues are means \pm SD. ^cValues are means \pm SD.

21. Quantity usually denotes the _____ **amount** of a substance expressed in grams or moles.
22. A hex-kinase is an enzyme found in 3 isoforms (HK1, 2 and 3), 1 type (HK1, 2 and 3), and 4 types (HK1, 2 and 3). "Purification" is "indicated separation" from the concentration of HK1, 2 and 3.

$$2000 + 2000 + 1 + 1 + 1 = 6003$$

4. Is the following sentence true or false? Sustainability is the satisfaction of present needs without compromising the ability of future generations to satisfy their own needs. **True**

85. *Cladonia* (liverworts) *Cladonia* against some mosses and fungi <http://www.mnh.si.edu/education/education.htm>

10. **What is the purpose of the "References" section in a research paper?**

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7. Consider the letters of the words that tell us what kind of information your **DATA SET** got from a chemical reaction.

- a. λ - wavelength
- b. ν - frequency
- c. n - index of refraction
- d. ν - number of particles
- e. λ - size of particles

Section 12 The Arithmetic Of Equations Practice Problems

B. T. P. S. Testing



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