

Understanding the Mole

Bean Lab (Relative Mass)

Introduction:

The relative mass of an object is how many times more massive the object is than a standard object. The atomic masses of atoms are all relative masses. They can be considered relative to any particular element. Historically, both oxygen and carbon have served as the reference standard. For our purposes we can also consider atomic masses relative to the least massive element--hydrogen, with an atomic mass of approx. one. Fluorine, with a relative mass of 19, is 19 times more massive than hydrogen, etc. In this laboratory exercise you deal with the relative masses of beans. Then you will be asked to draw a parallel to the atomic masses of elements.

Target:

Students are able to determine the relative mass of different small particles by comparison with an arbitrarily chosen mass. This will develop an understanding of the mole concept and molar masses of elements through an analogy with a model system.

Pre-lab assignment:

1. Read the entire lab
2. Print out the lab (data section to the end of the lab).
3. Below the target, answer the following questions regarding the lab:
 - a. Explain the process you will follow to figure out the mass of one bean.
 - b. Why should you calculate the mass of one bean rather than measuring one bean?
 - c. What should you record before you add the beans to the beaker?
 - d. In your own words, write down what is meant by "relative mass."

Procedure:

Part I. Be sure to use the same balance for the entire activity.

1. In a beaker, count out exactly 100 beans of one type. Discard any beans that differ greatly from an average bean. If you fail to do this, your results will not be accurate.
2. Put a beaker on the balance. Find the mass of the beaker. This you will need to subtract later. Add the 100 beans to the beaker; record their mass to the **nearest 0.01 g**.
3. Each team member is responsible for one type of bean provided.

Calculations:

1. Calculate (do not weigh) the mass of one bean of each type. Record the value in the data table. [Note: "Calculate" means to take the total mass of 100 beans and divide by 100 rather than weighing one particular bean.]
2. Determine the relative mass of each type of bean by comparison to the lightest type of bean. Record these values in the data table.

Relative	Mass	=	$\frac{\text{Avg. Mass of Bean}}{\text{Avg. Mass of Lightest Bean}}$
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3. Calculate the number of beans in one relative mass of each bean. Do this by dividing the relative mass by the average mass of one bean. Number of beans is = (one Relative mass/Average mass of one bean)

Understanding The Mole Bean Lab Answers

A Loxley



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Scientific and Technical Aerospace Reports ,1984 **Selected Water Resources Abstracts** ,1974 **Air Pollution Abstracts** ,1975 *Biological & Agricultural Index* ,1919 U.S. Government Research Reports ,1960 Resources for Teaching Middle School Science Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents Energy Research Abstracts ,1994 Semiannual with semiannual and annual indexes References to all scientific and technical literature coming from DOE its laboratories energy centers and contractors Includes all works deriving from DOE other related government sponsored information and foreign nonnuclear information Arranged under 39 categories e g Biomedical

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